

Computer Aided Drug Design Methods

Tugba ERTAN-BOLELLİ, Ph.D.

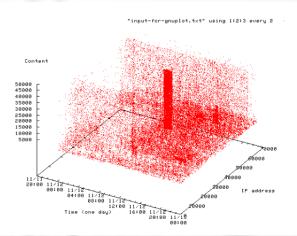
Associate Professor Ankara University, Faculty of Pharmacy, Pharmaceutical Chemistry Department



QUANTITATIVE STRUCTURE ACTIVITY RELATIONSHIPS (QSAR)

 QSAR methods are various forms of mathematical or statistical models that seek to predict the biological effects of chemicals based on their structure.

 Being able to predict biological activities of chemical structures by QSAR analysis dates back to the nineteenth century.



The aim of the QSAR is:

- to design a new compound that can shows better activity using the QSAR equation developed from a series of compounds,
- to reduce the toxicity of an existing compound,
- to optimize the hit compound (lead) with the optimum lipophilic property to pass a selected barrier (e.g. blood-brain barrier)

Biological Responses Used in QSAR Studies

- Affinity data: substrate or receptor binding
- Rate constants: association, dissociation
- Inhibition constants: IC50, enzyme inhibition values
- Pharmacokinetic parameters: absorption, distribution, metabolism, excretion
- In vitro and in vivo biological activity data
- Pharmacodynamic data of drugs (drug-receptor interaction)
- Toxic effect parameters

Physicochemical Parameters Used in QSAR Studies

PHYSICOCHEMICAL PARAMETERS	SYMBOL
LIPOPHILIC (HYDROPHOBIC) PARAMETERS	
Partition Coefficient	$Log P, (log P)^2$
π -Substituent Constant	$\pi, (\pi)^2$
Chromatography Distribution Coefficient (Liquid-liquid)	R _M
Hydrophobic Fragmental Constant	f
ELECTRONIC PARAMETERS	
Ionization Constant	pK _a
Sigma Aromatic Substituent Constant	$\sigma_{ m m}$, $\sigma_{ m m}$
Modification σ Aromatic Substituent Constants	$\sigma^+,\sigma^-,\sigma_1,\sigma_R,\sigma^o$
Sigma Aliphatic Substituent Constant	σ*
Substituent Resonance Effect	R
Substituent Inductive Effect	F
QUANTUM MECHANICAL PARAMETERS	
Atomic σ Elektron Charge	q^{σ}, Q^{σ}
Atomic π Elektron Charge	q^{π}, Q^{π}
Nucleophilic Delocalization State	S _r ^N
Electrophilic Delocalization State	S _r ^E
Energy of Lowest Unoccupied Molecular Orbital, "electrophilicity" Energy of Highest Occupied Molecular Orbital, "nucleophilicity"	E _{LUMO}
	E _{HOMO}
STERIC PARAMETERS	
Steric Substituent Constant	Es
Molar Volume	MV
Molar Refractivity Substituent Constant	MR
Molecular Weight	MW
Van der Waals Radii Sterimol Width and Length Parameters	R
Stermor width and Length Faralleters	L, B ₁ -B ₄



Lipophilic Property

- is the most used physicochemical property in QSAR studies.
- Lipophilicity can be defined as the dispersion between water and oil phase.

Log P = Partition Coefficient

- It is a parameter that expresses the concentration of the chemical compound distributed between the lipid-water layers. For this purpose, it was found that the most suitable solvent system is 1-octanol / water.
- As the water, the buffer solution is prepared to mimic the physiological pH (pH = 7.4).

Structure-Activity Relationships (QSAR) Analysis

In the 1960s, two different QSAR analysis methods were developed. They were developed by

- Hansch and Fujita,
- Free and Wilson.

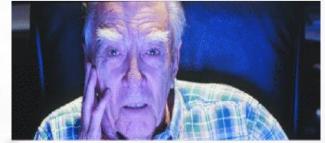
Quantitative structure-activity relationships (QSAR) are the mathematical methods for describing the relationships between molecular properties of chemical compounds (structural / physicochemical properties) and biological activities.



Tuaba ERTAN-BOLELLI

Hansch Analysis Method

In the analysis method, Hansch expressed that the observed biological effects of the compounds in a homologous series are a function of the physicochemical properties of these compounds and developed the following formula;



Hansch

biological effect = f (hydrophobic) + f (electronic) + f (steric) + c (constant)



biological effect = f (hydrophobic) + f (electronic) + f (steric) + c (constant)

Log 1 / C = Logarithmic biological effect Independent variables of physicochemical parameters

Y (biological activity) = $k_0 + k_1 X_1 + k_2 X_2 + \dots + k_n X_n$

correlation constant indicating the contribution of the unexplained residue to the biological activity

The constants (regression coefficients) that define (+) or (-) contribution of physicochemical properties to biological activity

Hansch equations

$$log 1/C = 1.22 \pi - 1.59 \sigma + 7.89$$

(n=22; s=0.238; r= 0.918
log 1/C = 0.398 \pi + 1.089 \sigma + 1.03 Es + 4.541
(n=9; r= 0.955)

