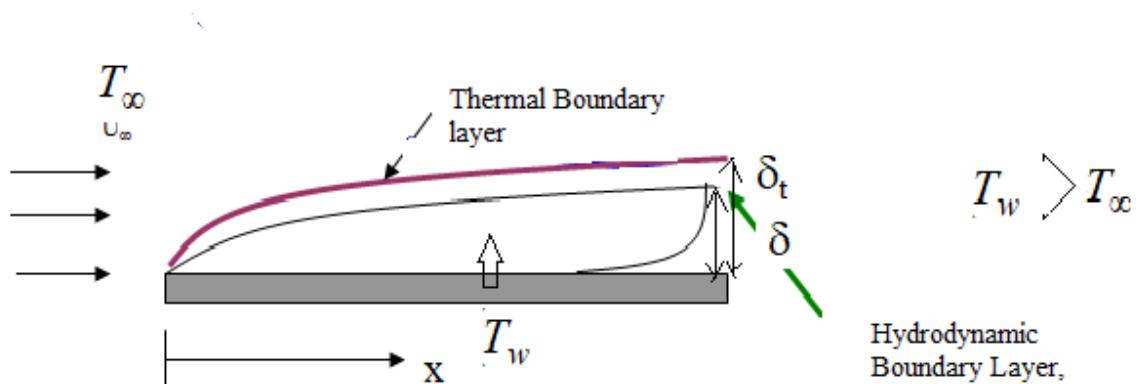


References:

- W.L. McCabe, J.C. Smith, P. Harriott., *Unit Operations of Chemical Engineering*, McGraw Hill, N.Y., (7th Ed.) 2005
- J.P. Holman, *Heat Transfer*, McGraw-Hill, N.Y., 1989.
- F.P. Incropera, D.P. de Witt, *Fundamentals of Heat and Mass Transfer*, John Wiley & Sons, N.Y., (3th Ed.) 1990.
- C.J. Geankoplis, *Transport Processes and Unit Operations*, Prentice-Hill Inc., N.J., (3th Ed.) 1993
- Y. Cengel , *Introduction to Thermodynamics and Heat Transfer* , , McGraw Hill, 2nd Edition 2008

5. Heat transfer coefficients, thermal boundary layer



$$q = -kA\left(\frac{\partial T}{\partial y}\right)_{y=0} = Ah(T_w - T_\infty)$$

Laminar flow heat transfer to flat plate

$$\overline{Nu} = 0.664 \text{Re}^{1/2} \text{Pr}^{1/3}$$

Example:

$$T_\infty = 20^\circ\text{C}$$

$$v = 15 \text{ m/s}$$

$$T_w = 110^\circ\text{C}$$

$$x = 0.5 \text{ m}$$

$$q = ?$$

$$q = h A (T_w - T_\infty)$$

$$T_f = \frac{T_w + T_\infty}{2} = \frac{110 + 20}{2} = 65^\circ\text{C} = 338\text{ K}$$

Physical properties of air at 338 K (Table 5) by interpolation,

$$\begin{aligned}\rho &= 1.041 \text{ kg/m}^3 \\ C_p &= 1.0082 \text{ kJ/kg}^\circ\text{C} \\ \mu &= 2.02 * 10^{-5} \text{ kg/ms} \\ k &= 0.02912 \text{ W/m}^\circ\text{C} \\ Pr &= 0.699\end{aligned}$$

$$Re = \frac{x v \rho}{\mu} = \frac{0.5 \text{ m} * 15 \text{ m/s} * 1.041 \text{ kg/m}^3}{2.02 * 10^{-5} \text{ kg/ms}} = 386509 < 5 * 10^5 \quad (\text{Laminar flow})$$

$$Nu = 0.664 * Pr^{1/3} * Re^{1/2}$$

$$Nu = 0.664 * (0.699)^{1/3} * (386509)^{1/2} = 366.36$$

$$Nu = \frac{h x}{k} = 366.36 = \frac{h * 0.5 \text{ m}}{0.02912 \text{ W/m}^\circ\text{C}}$$

$$h = 21.34 \text{ W/m}^2{}^\circ\text{C}$$

$$q = h A (T_w - T_\infty)$$

$$q = 21.34 * 0.5^2 * (110 - 20)$$

$$q = 480.15 \text{ W}$$