

References:

- W.L. McCabe, J.C. Smith, P. Harriott., *Unit Operations of Chemical Engineering*, McGraw Hill, N.Y., (7th Ed.) 2005
- J.P. Holman, *Heat Transfer*, McGraw-Hill, N.Y., 1989.
- F.P. Incropera, D.P. de Witt, *Fundamentals of Heat and Mass Transfer*, John Wiley & Sons, N.Y., (3th Ed.) 1990.
- C.J. Geankoplis, *Transport Processes and Unit Operations*, Prentice-Hill Inc., N.J., (3th Ed.) 1993
- Y. Cengel , *Introduction to Thermodynamics and Heat Transfer* , , McGraw Hill, 2nd Edition 2008

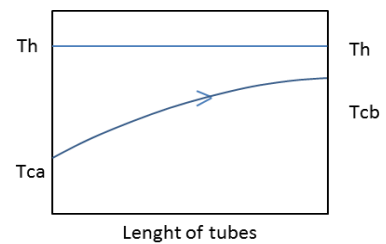
6. Energy balances in heat exchangers

- a) **Hot fluid:** condensing vapour (enters condenser as saturated vapour and the condensate leaves at condensing temperature (T_h) without any further cooled)

Cold fluid: cooling water

Countercurrent flow heat exchanger

$$q = m \lambda = m_c C_{pc} (T_{ca} - T_{cb})$$

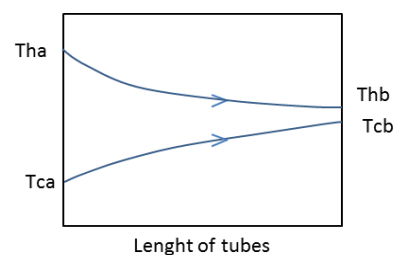


- b) **Hot fluid:** aniline

Cold fluid: toluene

Parallel flow heat exchanger

$$q = m_c C_{pc} (T_{ca} - T_{cb}) = m_h C_{ph} (T_{ha} - T_{hb})$$



- c) **Hot fluid:** superheated vapor

Cold fluid: cooling water

Countercurrent flow heat exchanger

$$q = m \lambda + m_h C_{ph} (T_{ha} - T_h) = m_c C_{pc} (T_{ca} - T_{cb})$$

