

Ankara University, Faculty of Agriculture , Department of Fisheries and Aquaculture, Programme of Fisheries and Aquaculture

AQS104: Biochemistry

Reference: Nelson, D. L., Lehninger, A. L., & Cox, M. M. (2008). ***Lehninger Principles of Biochemistry (5th edition)***. Macmillan.

AQS104 BIOCHEMISTRY: Weekly Programme

1. Week:

- The foundations of biochemistry
- Water

2. Week:

- Amino acids, peptides, and proteins
- The three-dimensional structure of proteins

3. Week:

- Protein function
- Enzymes

4. Week:

- Carbohydrates and Glycobiology
- Nucleotides and Nucleic Acids

5. Week:

- DNA-based information technologies
- Lipids

6. Week:

Biological membranes and transport
Biosignaling

7. Week:

Bioenergetics and biochemical reaction types
Glycolysis, gluconeogenesis, and the pentose phosphate pathway

8. Week:

Principles of metabolic regulation
The citric acid cycle

9. Week:

Fatty acid catabolism
Aino acid oxidation and the production of urea

10. Week:

Oxidative phosphorylation and photophosphorylation
Carbohydrate biosynthesis in plants and bacteria

11. Week:

Lipid biosynthesis
Biosynthesis of amino acids, nucleotides, and related molecules

12. Week:

Hormonal regulation and integration of mammalian metabolism
Genes and chromosomes

13. Week:

DNA metabolism
RNA metabolism

14. Week:

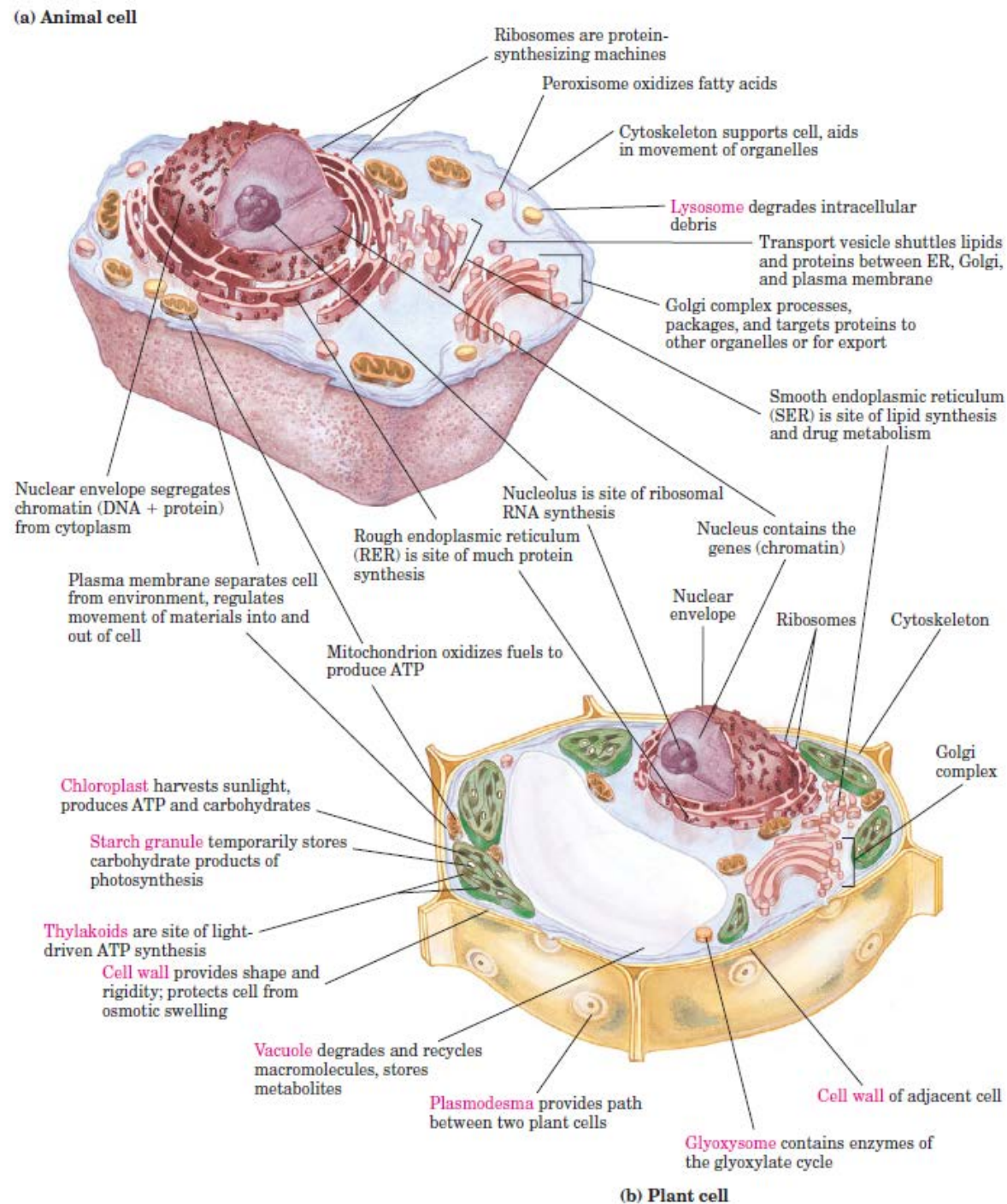
Protein metabolism
Regulation of gene expression

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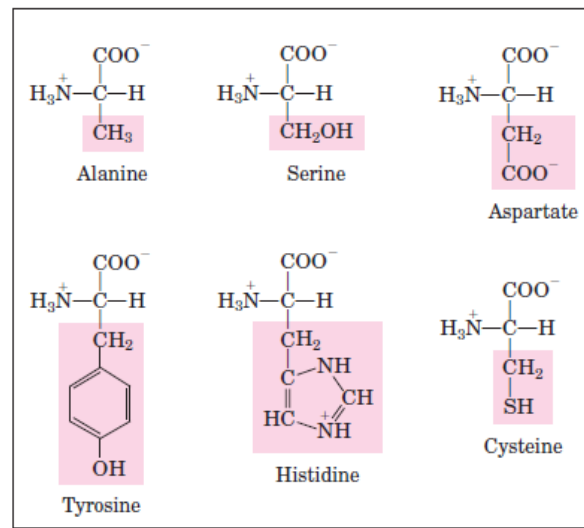
AQS104: Biochemistry

1. Week: The Foundations of Biochemistry Water

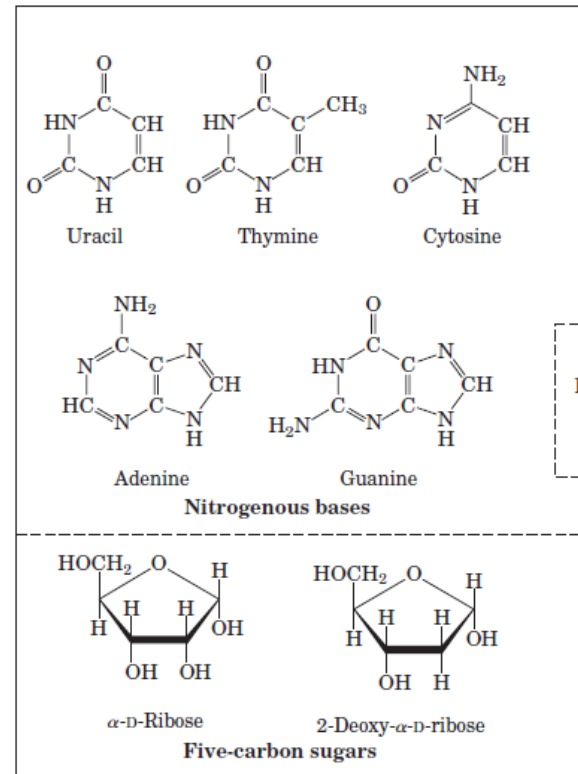
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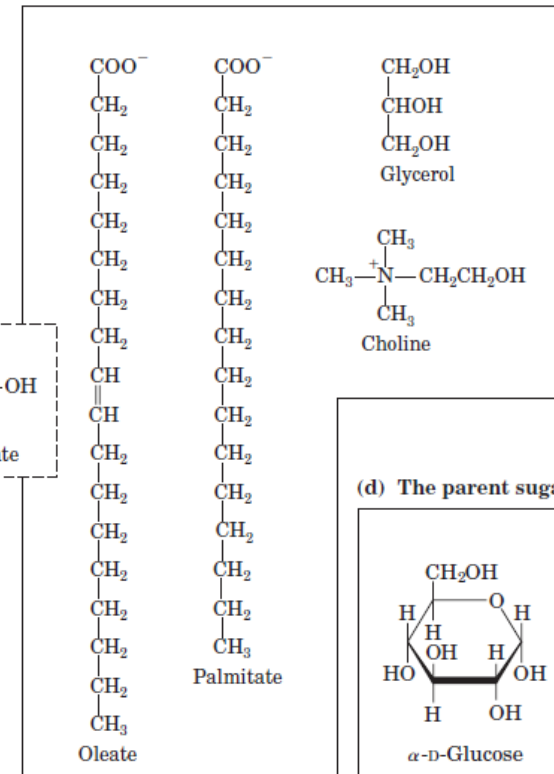
(a) Some of the amino acids of proteins



(b) The components of nucleic acids



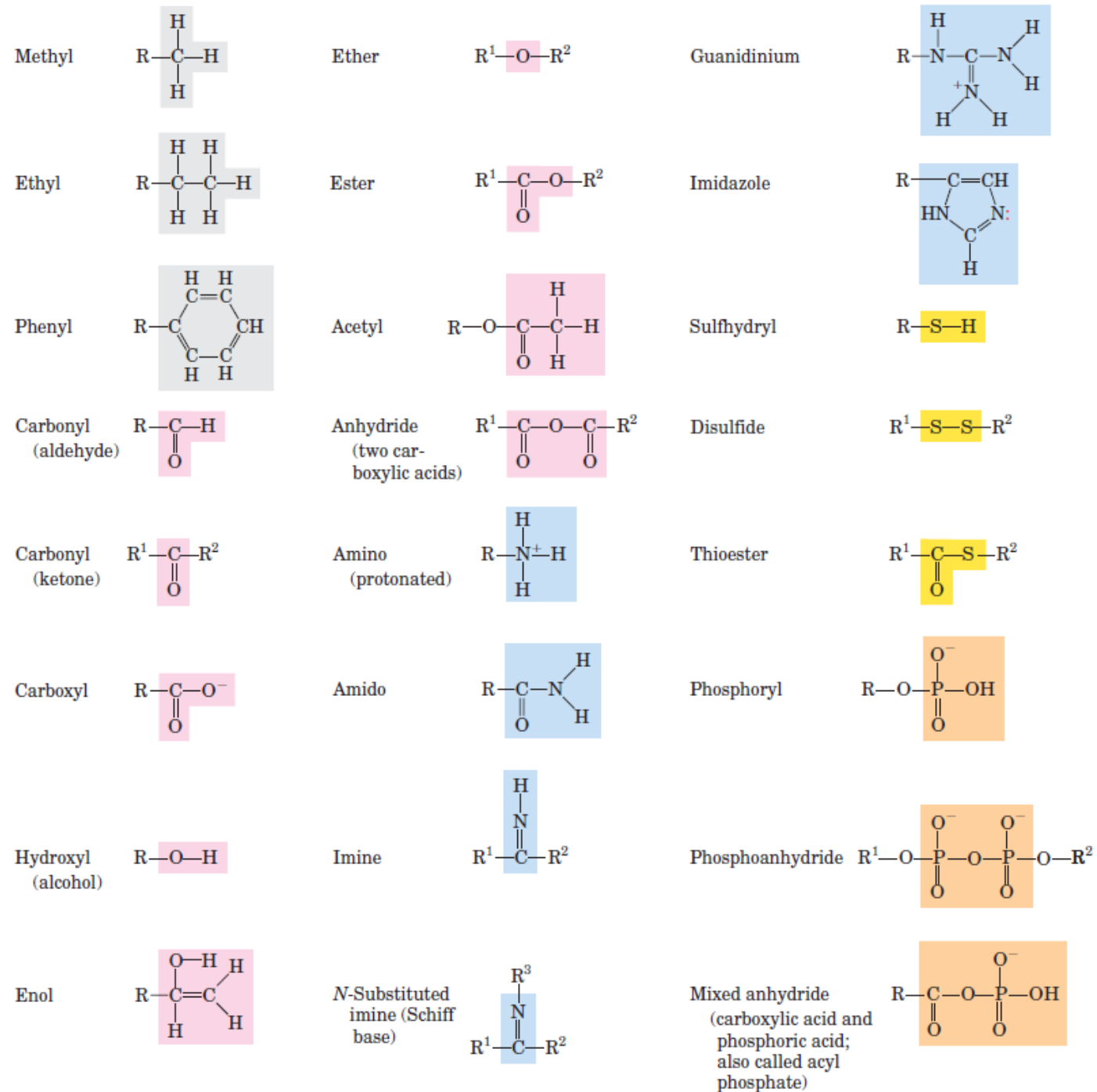
(c) Some components of lipids

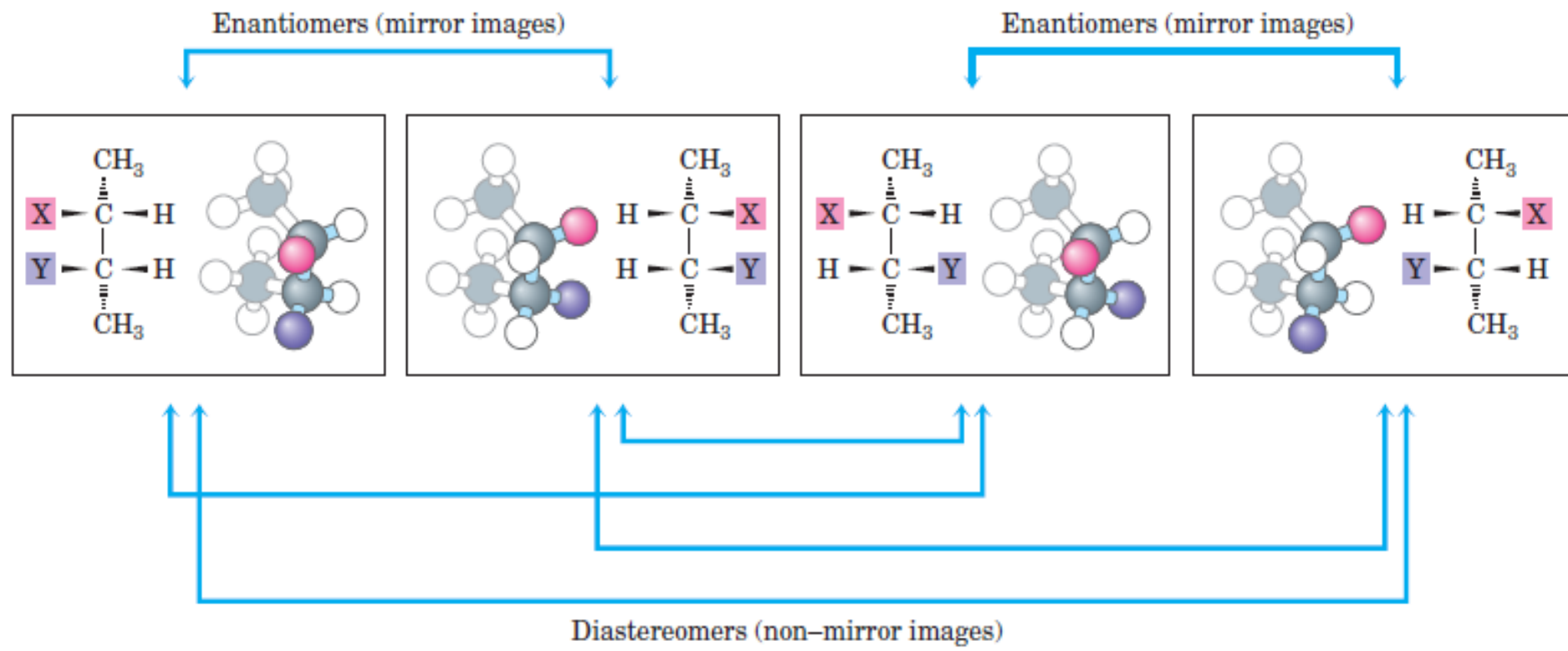


Figures & Tables are taken from: Nelson, D. L., Lehninger, A. L., & Cox, M. M. (2008). *Lehninger Principles of Biochemistry (5th edition)*. Macmillan.

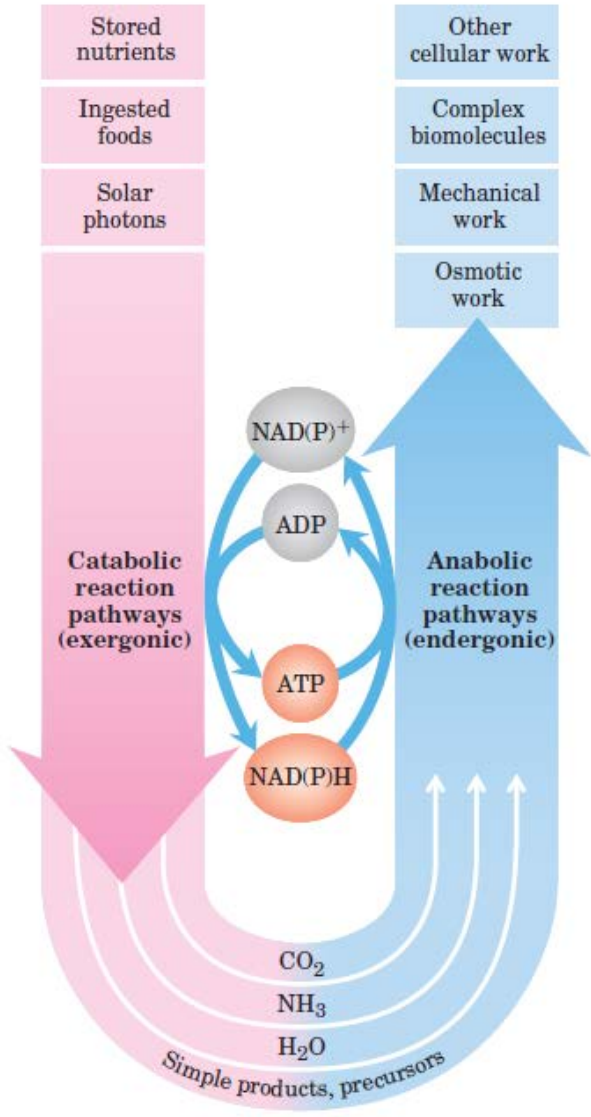
1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba		72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra		<div style="display: flex; justify-content: space-between; align-items: center;"> <div style="text-align: center;"> <p>Lanthanides</p> <p>Actinides</p> </div> <div style="text-align: center;"> <p>←</p> <p>←</p> </div> </div>														

Bulk elements
 Trace elements





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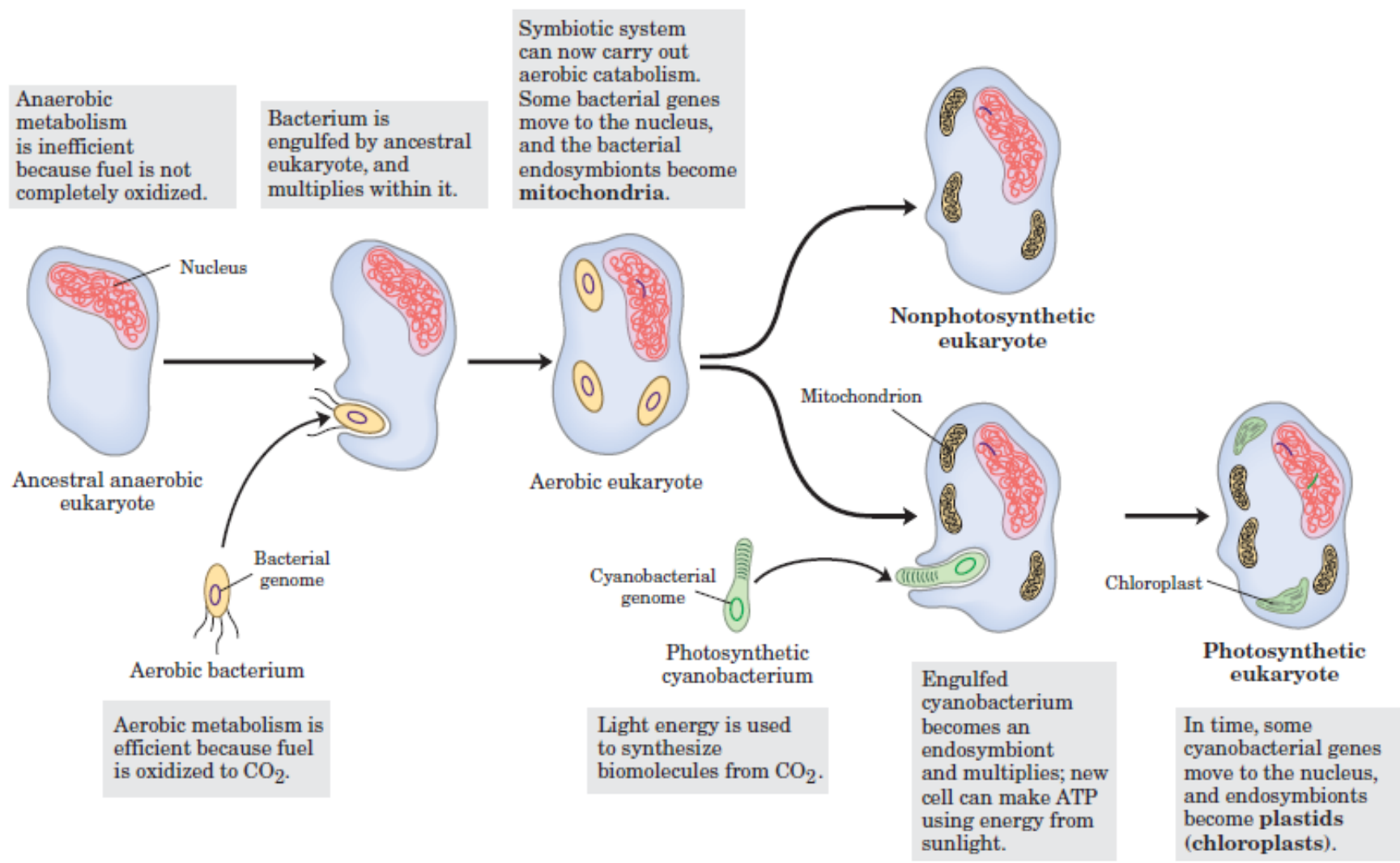
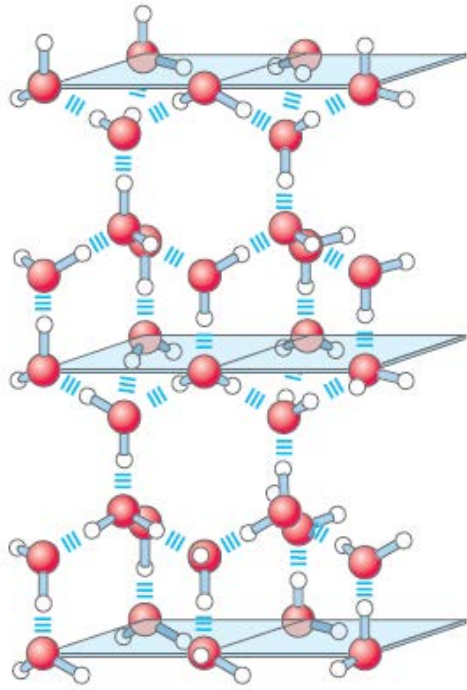
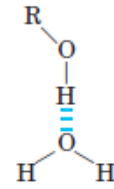


TABLE 1–2 A Few of the Many Organisms Whose Genomes Have Been Completely Sequenced			
Organism	Genome size (nucleotide pairs)	Number of genes	Biological interest
<i>Mycoplasma genitalium</i>	5.8×10^5	4.8×10^2	Smallest true organism
<i>Treponema pallidum</i>	1.1×10^6	1.0×10^3	Causes syphilis
<i>Borrelia burgdorferi</i>	9.1×10^5	8.5×10^2	Causes Lyme disease
<i>Helicobacter pylori</i>	1.7×10^6	1.6×10^3	Causes gastric ulcers
<i>Methanococcus jannaschii</i>	1.7×10^6	1.7×10^3	Archaeon; grows at 85 °C!
<i>Haemophilus influenzae</i>	1.8×10^6	1.6×10^3	Causes bacterial influenza
<i>Archaeoglobus fulgidus*</i>	2.2×10^6	2.4×10^3	High-temperature methanogen
<i>Synechocystis</i> sp.	3.6×10^6	3.2×10^3	Cyanobacterium
<i>Bacillus subtilis</i>	4.2×10^6	4.1×10^3	Common soil bacterium
<i>Escherichia coli</i>	4.6×10^6	4.4×10^3	Some strains are human pathogens
<i>Saccharomyces cerevisiae</i>	1.2×10^7	5.9×10^3	Unicellular eukaryote
<i>Plasmodium falciparum</i>	2.3×10^7	5.3×10^3	Causes human malaria
<i>Caenorhabditis elegans</i>	1.0×10^8	2.3×10^4	Multicellular roundworm
<i>Anopheles gambiae</i>	2.3×10^8	1.3×10^4	Malaria vector
<i>Arabidopsis thaliana</i>	1.2×10^8	3.2×10^4	Model plant
<i>Oryza sativa</i>	3.9×10^8	3.8×10^4	Rice
<i>Drosophila melanogaster</i>	1.2×10^8	2.0×10^4	Laboratory fly (“fruit fly”)
<i>Mus musculus domesticus</i>	2.6×10^9	2.7×10^4	Laboratory mouse
<i>Pan troglodytes</i>	3.1×10^9	4.9×10^4	Chimpanzee
<i>Homo sapiens</i>	3.1×10^9	2.9×10^4	Human

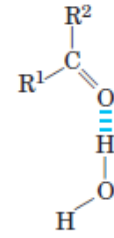
Source: RefSeq page for each organism at www.ncbi.nlm.nih.gov/genomes.



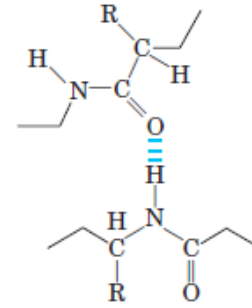
Between the hydroxyl group of an alcohol and water



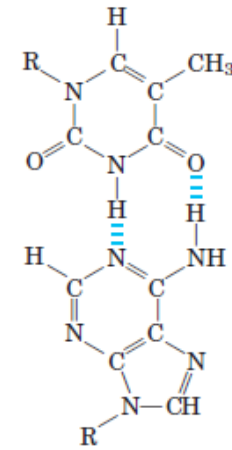
Between the carbonyl group of a ketone and water



Between peptide groups in polypeptides

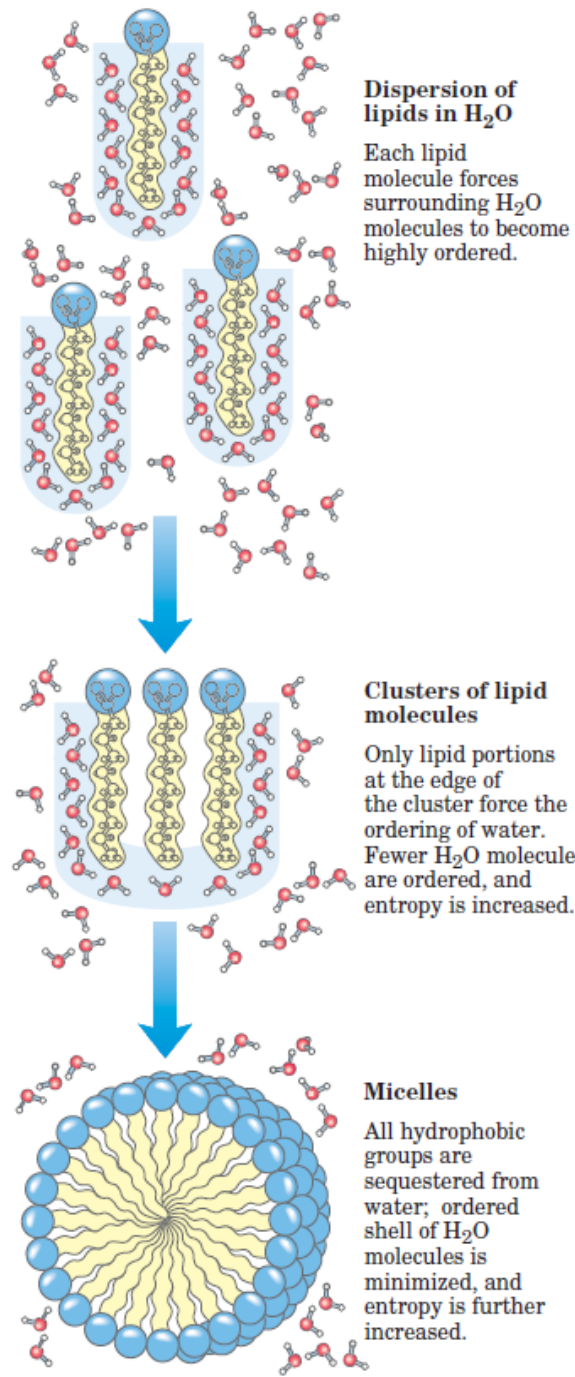
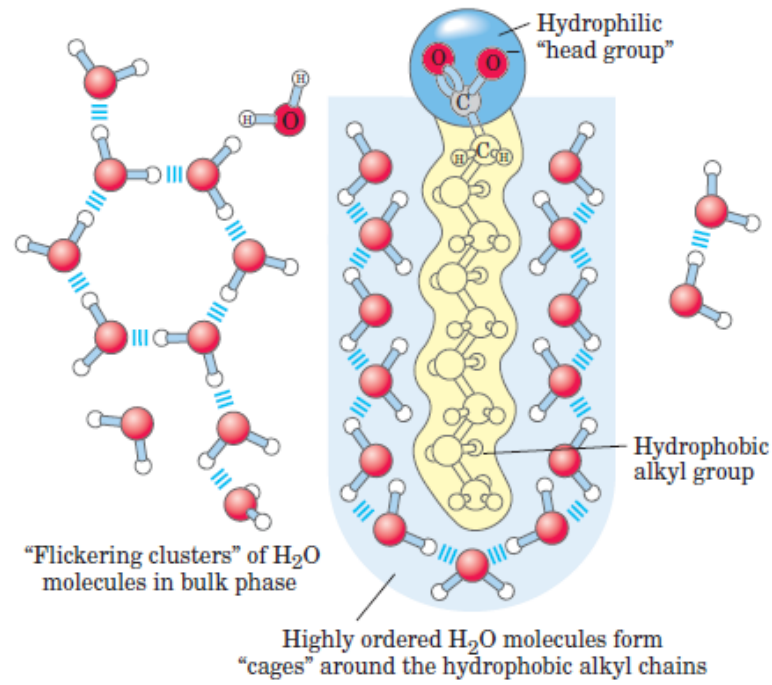


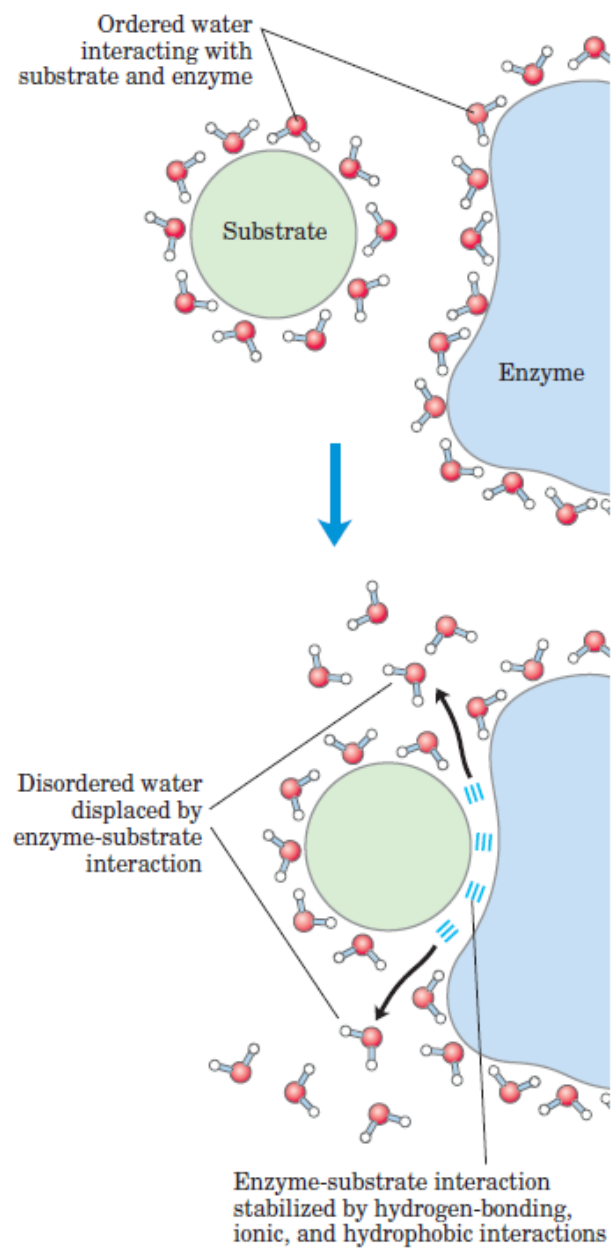
Between complementary bases of DNA



Thymine

Adenine





Monoprotic acids

Acetic acid
($K_a = 1.74 \times 10^{-5} \text{ M}$)

Ammonium ion
($K_a = 5.62 \times 10^{-10} \text{ M}$)

Diprotic acids

Carbonic acid
($K_a = 1.70 \times 10^{-4} \text{ M}$);
Bicarbonate
($K_a = 6.31 \times 10^{-11} \text{ M}$)

Glycine, carboxyl
($K_a = 4.57 \times 10^{-3} \text{ M}$);
Glycine, amino
($K_a = 2.51 \times 10^{-10} \text{ M}$)

Triprotic acids

Phosphoric acid
($K_a = 7.25 \times 10^{-3} \text{ M}$);
Dihydrogen phosphate
($K_a = 1.38 \times 10^{-7} \text{ M}$);
Monohydrogen phosphate
($K_a = 3.98 \times 10^{-13} \text{ M}$)

