

Ankara University, Faculty of Agriculture , Department of Fisheries and Aquaculture, Programme of Fisheries and Aquaculture

# AQS104: Biochemistry

Reference: Nelson, D. L., Lehninger, A. L., & Cox, M. M. (2008). ***Lehninger Principles of Biochemistry (5<sup>th</sup> edition)***. Macmillan.

# AQS104 BIOCHEMISTRY: Weekly Programme

**1. Week:**

- The foundations of biochemistry
- Water

**2. Week:**

- Amino acids, peptides, and proteins
- The three-dimensional structure of proteins

**3. Week:**

- Protein function
- Enzymes

**4. Week:**

- Carbohydrates and Glycobiology
- Nucleotides and Nucleic Acids

**5. Week:**

- DNA-based information technologies
- Lipids

**6. Week:**

Biological membranes and transport  
Biosignaling

**7. Week:**

Bioenergetics and biochemical reaction types  
Glycolysis, gluconeogenesis, and the pentose phosphate pathway

**8. Week:**

Principles of metabolic regulation  
The citric acid cycle

**9. Week:**

Fatty acid catabolism  
Aino acid oxidation and the production of urea

**10. Week:**

Oxidative phosphorylation and photophosphorylation  
Carbohydrate biosynthesis in plants and bacteria

**11. Week:**

Lipid biosynthesis  
Biosynthesis of amino acids, nucleotides, and related molecules

**12. Week:**

Hormonal regulation and integration of mammalian metabolism  
Genes and chromosomes

**13. Week:**

DNA metabolism  
RNA metabolism

**14. Week:**

Protein metabolism  
Regulation of gene expression

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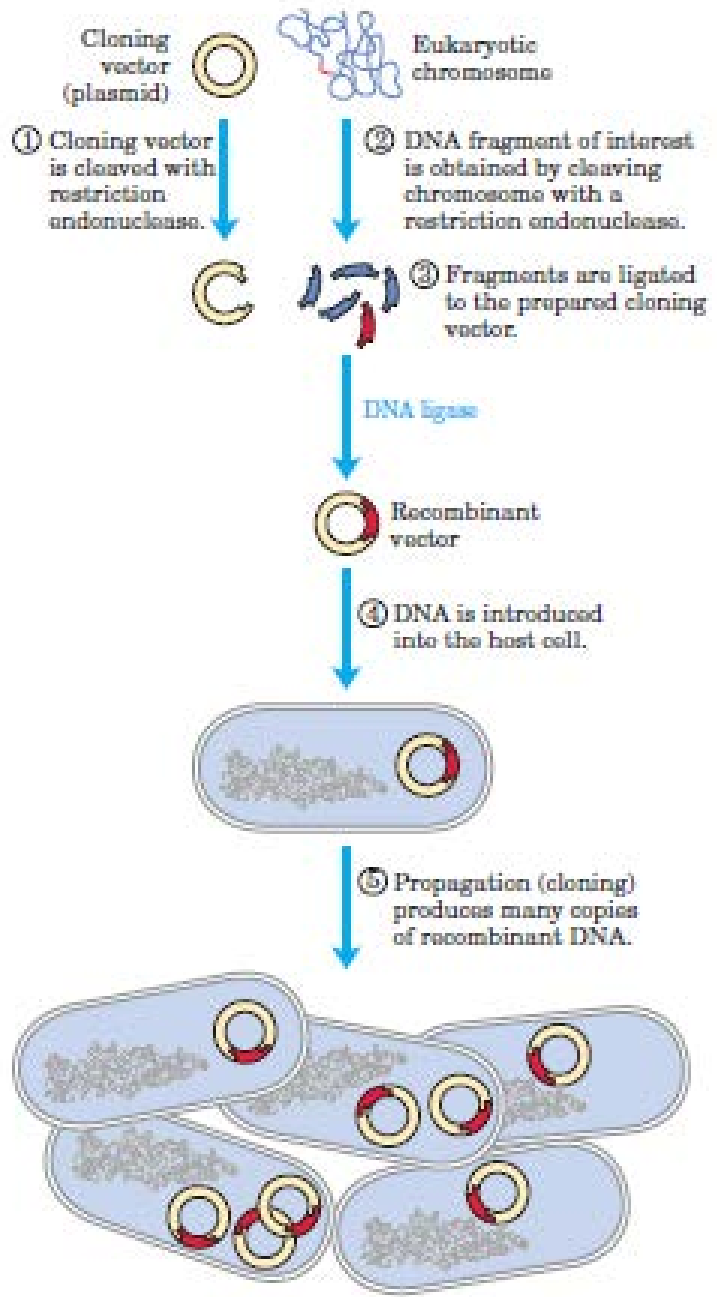
### 5. Week:

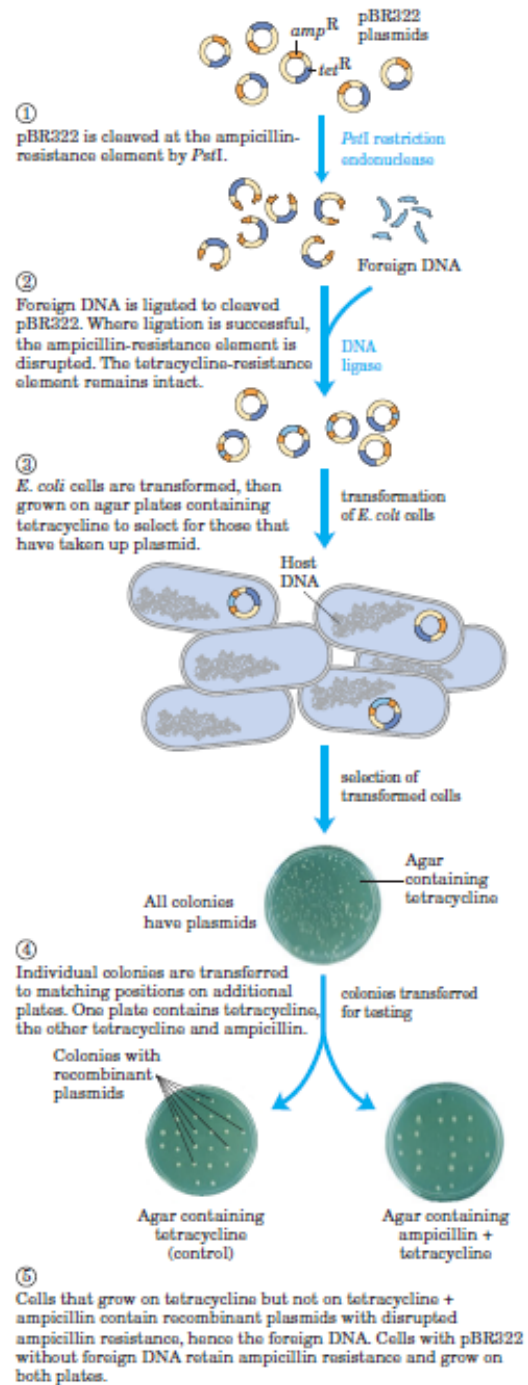
## DNA-based Information Technologies

### Lipids

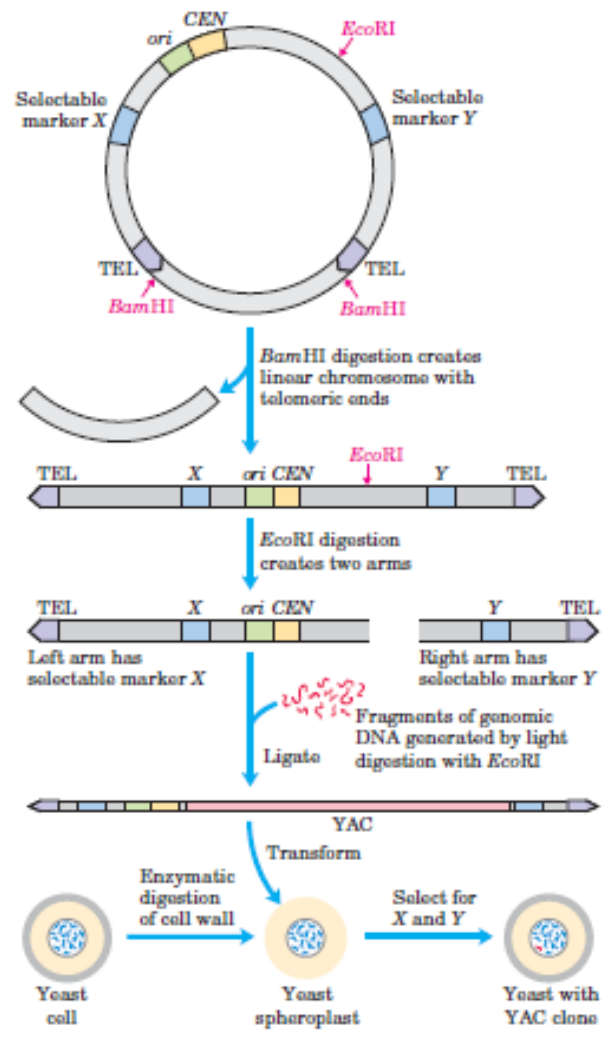
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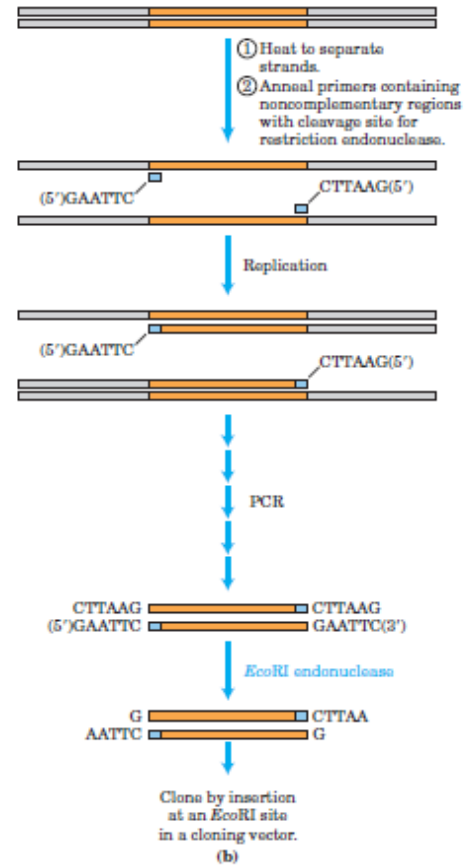
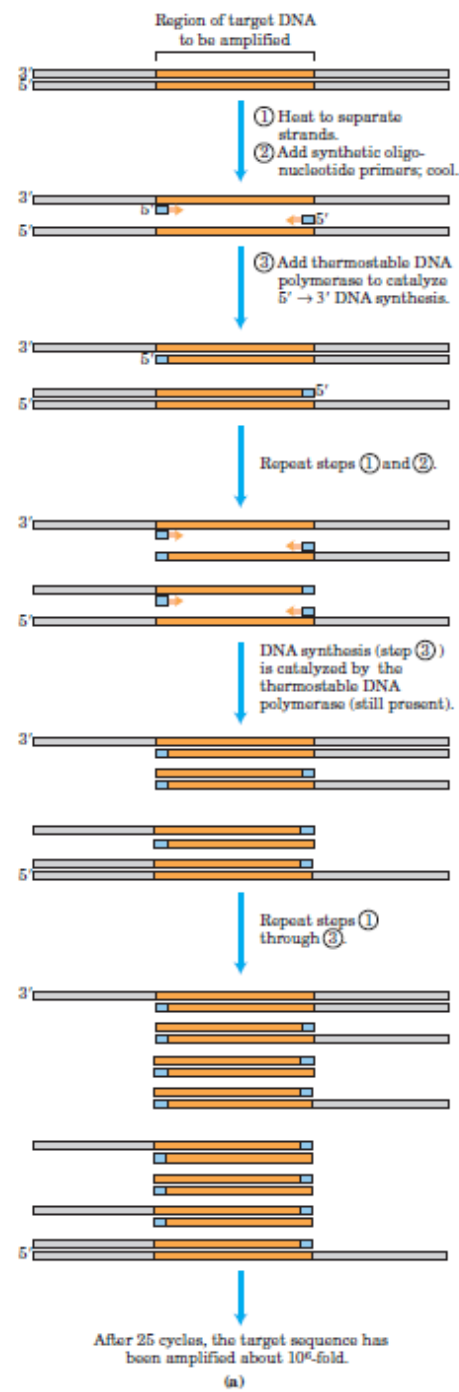
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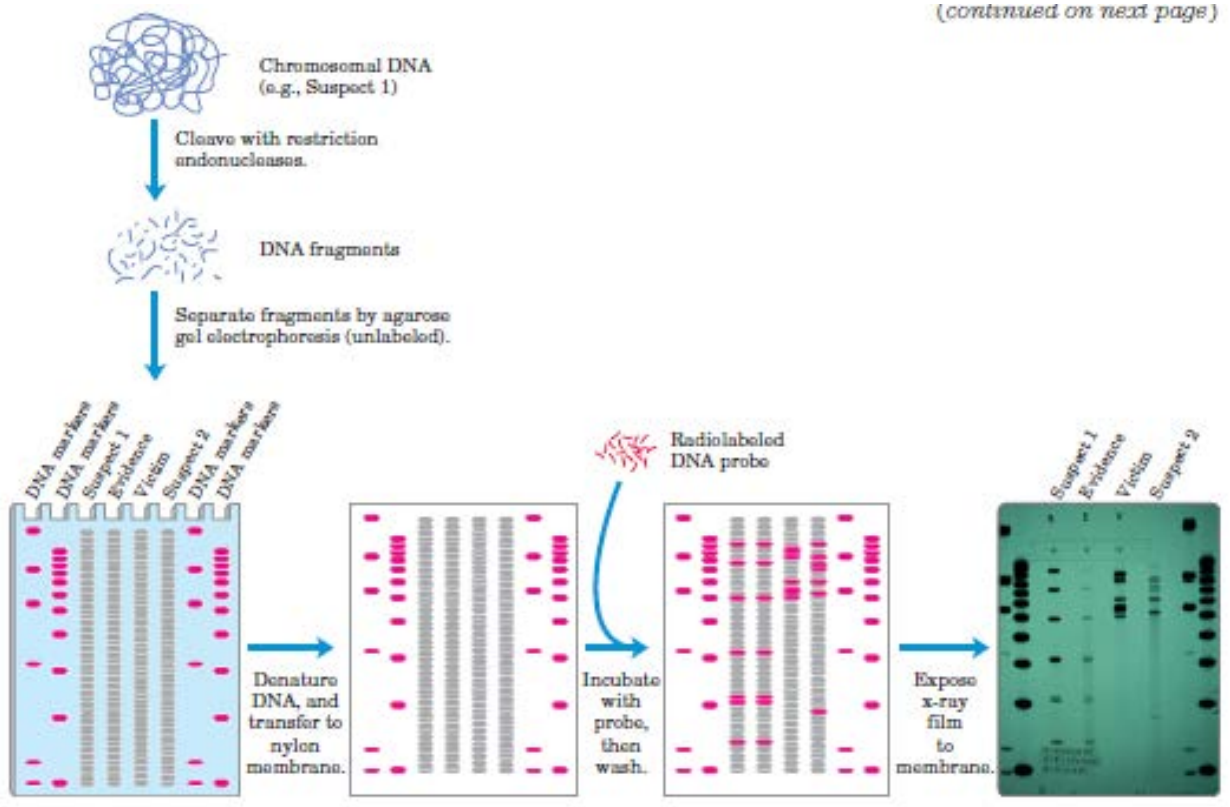




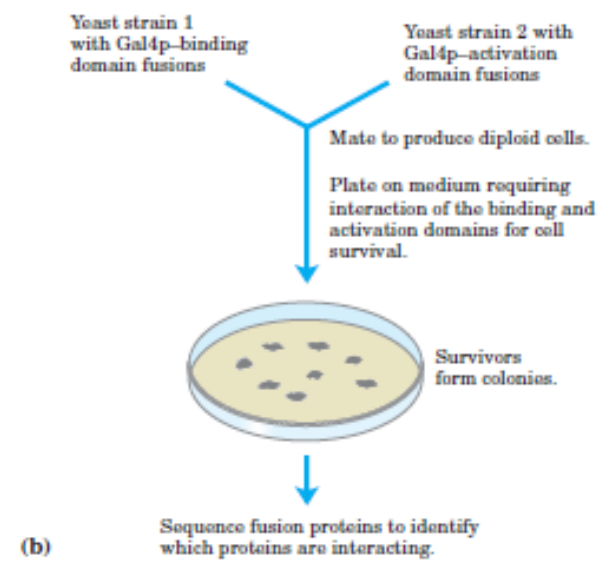
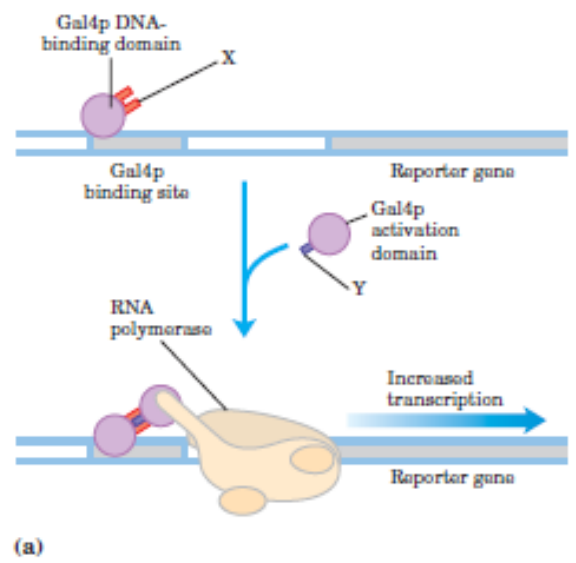
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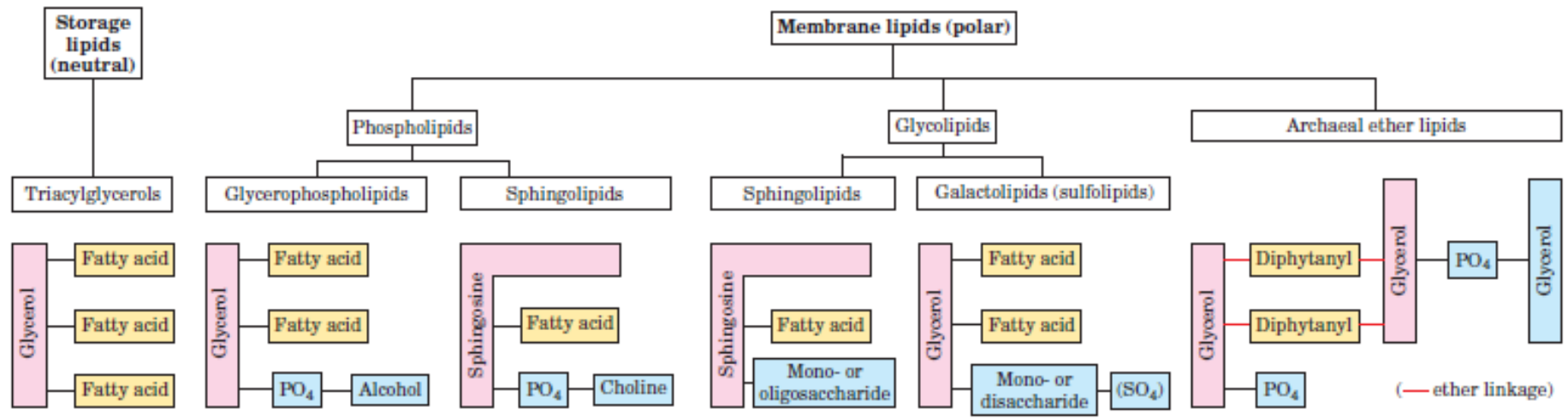


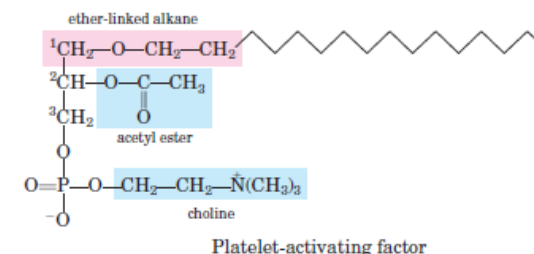
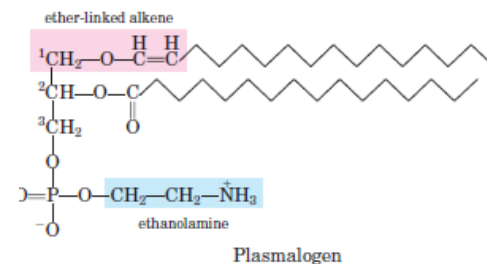
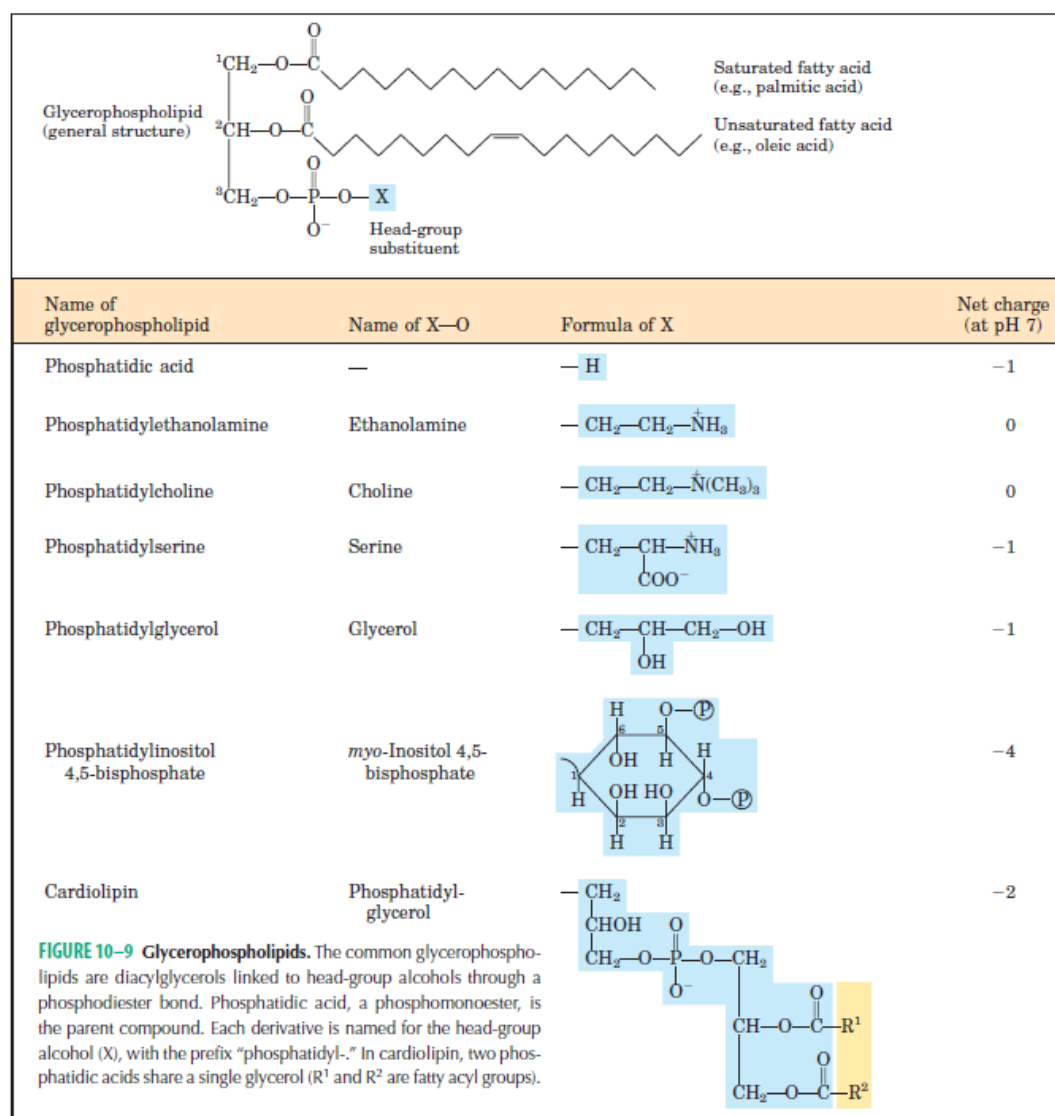
Carbon skeleton	Structure*	Systematic name <sup>†</sup>	Common name (derivation)	Melting point (°C)	Solubility at 30 °C (mg/g solvent)	
					Water	Benzene
12:0	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>10</sub> COOH	<i>n</i> -Dodecanotic acid	Lauric acid (Latin <i>laurus</i> , "laurel plant")	44.2	0.063	2,600
14:0	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>12</sub> COOH	<i>n</i> -Tetradecanotic acid	Myristic acid (Latin <i>Myristica</i> , nutmeg genus)	53.9	0.024	874
16:0	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>14</sub> COOH	<i>n</i> -Hexadecanotic acid	Palmitic acid (Latin <i>palma</i> , "palm tree")	63.1	0.0083	348
18:0	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>16</sub> COOH	<i>n</i> -Octadecanotic acid	Stearic acid (Greek <i>stear</i> , "hard fat")	69.6	0.0034	124
20:0	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>18</sub> COOH	<i>n</i> -Eicosanotic acid	Arachidic acid (Latin <i>Arachis</i> , legume genus)	76.5		
24:0	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>22</sub> COOH	<i>n</i> -Tetracosanotic acid	Lignoceric acid (Latin <i>lignum</i> , "wood" + <i>ceru</i> , "wax")	86.0		
16:1(Δ <sup>9</sup> )	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>5</sub> CH=CH(CH <sub>2</sub> ) <sub>7</sub> COOH	<i>cis</i> -9-Hexadecenotic acid	Palmitoleic acid	1 to -0.5		
18:1(Δ <sup>9</sup> )	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>7</sub> CH=CH(CH <sub>2</sub> ) <sub>7</sub> COOH	<i>cis</i> -9-Octadecenotic acid	Oleic acid (Latin <i>oleum</i> , "oil")	13.4		
18:2(Δ <sup>9,12</sup> )	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>4</sub> CH=CHCH <sub>2</sub> CH=CH(CH <sub>2</sub> ) <sub>7</sub> COOH	<i>cis-cis</i> -9,12-Octadecadienotic acid	Linoleic acid (Greek <i>linon</i> , "flax")	1-5		
18:3(Δ <sup>9,12,15</sup> )	CH <sub>3</sub> CH <sub>2</sub> CH=CHCH <sub>2</sub> CH=CHCH <sub>2</sub> CH=CH(CH <sub>2</sub> ) <sub>7</sub> COOH	<i>cis-cis-cis</i> -9,12,15-Octadecatrienotic acid	α-Linolenic acid	-11		
20:4(Δ <sup>5,8,11,14</sup> )	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>4</sub> CH=CHCH <sub>2</sub> CH=CHCH <sub>2</sub> CH=CHCH <sub>2</sub> CH=CH(CH <sub>2</sub> ) <sub>5</sub> COOH	<i>cis-cis-cis-cis</i> -5,8,11,14-Icosatetraenotic acid	Arachidonic acid	-49.5		

\*All acids are shown in their nonionized form. At pH 7, all free fatty acids have an ionized carboxylate. Note that numbering of carbon atoms begins at the carboxyl carbon.

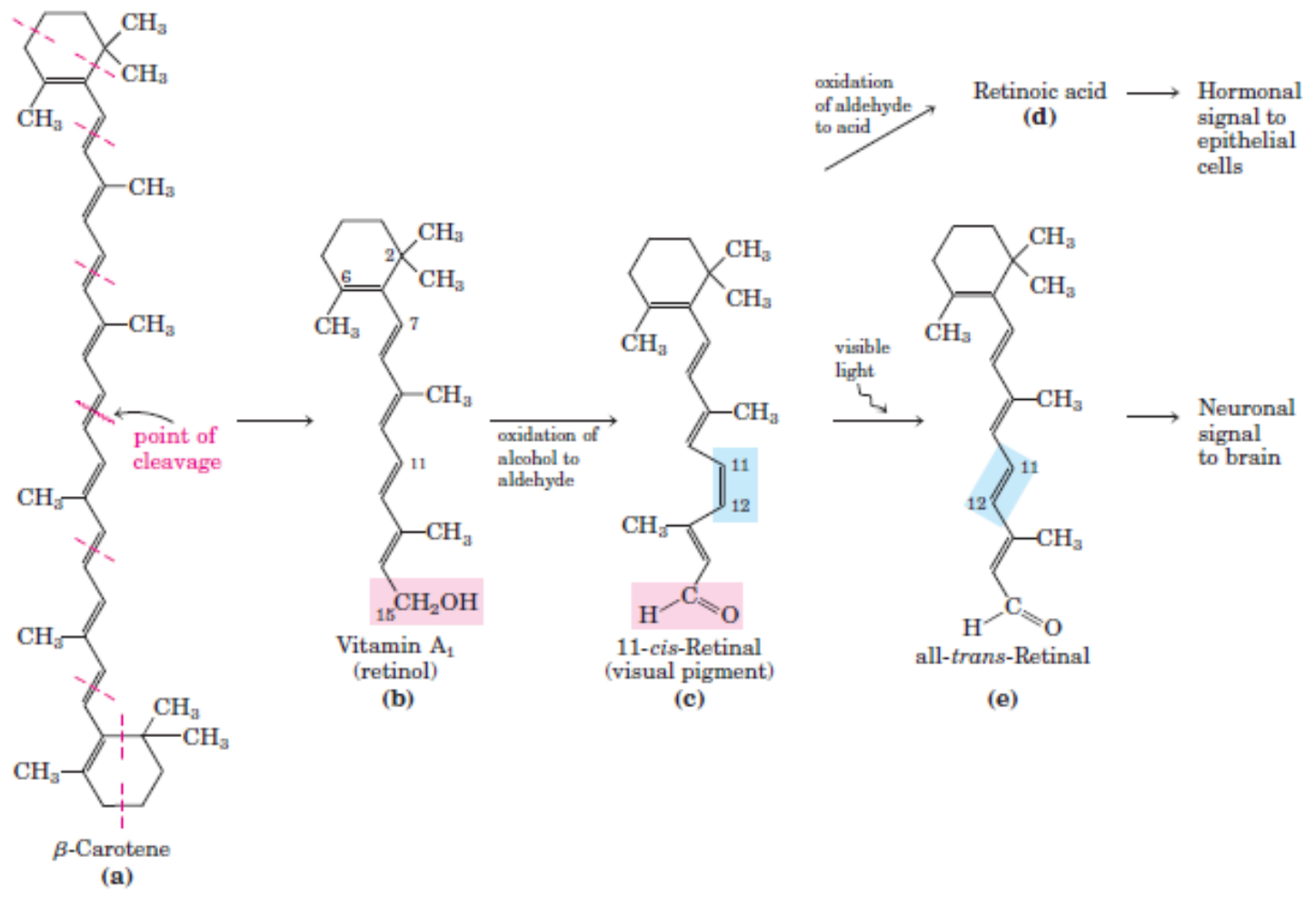
<sup>†</sup>The prefix *n*- indicates the "normal" unbranched structure. For instance, "dodecanotic" simply indicates 12 carbon atoms, which could be arranged in a variety of branched forms; "*n*-dodecanotic" specifies the linear, unbranched form. For unsaturated fatty acids, the configuration of each double bond is indicated; in biological fatty acids the configuration is almost always *cis*.

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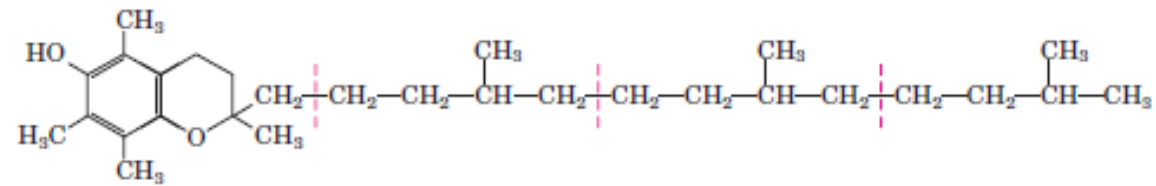




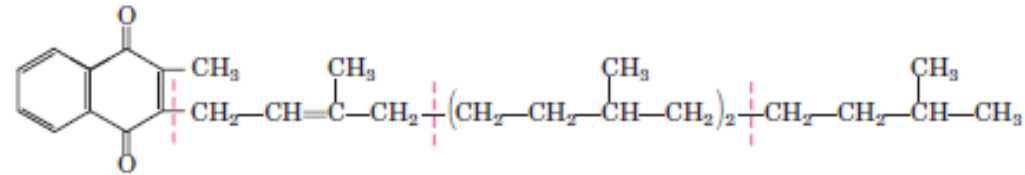
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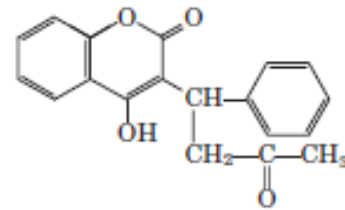
**(a)**  
Vitamin E: an antioxidant



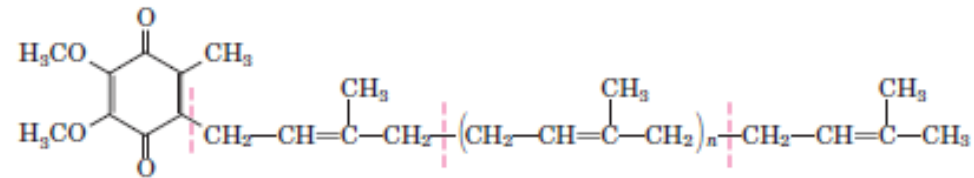
**(b)**  
Vitamin K<sub>1</sub>: a blood-clotting cofactor (phylloquinone)



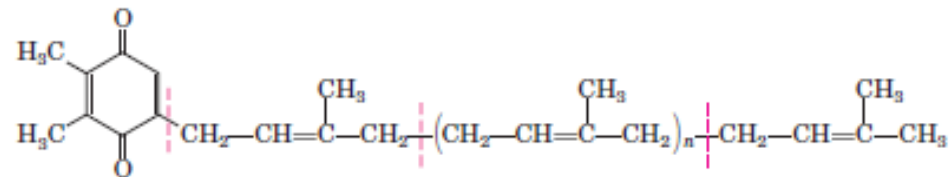
**(c)**  
Warfarin: a blood anticoagulant



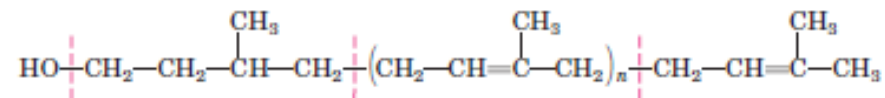
**(d)**  
Ubiquinone: a mitochondrial electron carrier (coenzyme Q)  
( $n = 4$  to  $8$ )



**(e)**  
Plastoquinone: a chloroplast electron carrier ( $n = 4$  to  $8$ )



**(f)**  
Dolichol: a sugar carrier  
( $n = 9$  to  $22$ )



Category	Category code	Examples
Fatty acids	FA	Oleate, stearyl-CoA, palmitoylcarnitine
Glycerolipids	GL	Di- and triacylglycerols
Glycerophospholipids	GP	Phosphatidylcholine, phosphatidylserine, phosphatidylethanolamine
Sphingolipids	SP	Sphingomyelin, ganglioside GM2
Sterol lipids	ST	Cholesterol, progesterone, bile acids
Prenol lipids	PR	Farnesol, geraniol, retinol, ubiquinone
Saccharolipids	SL	Lipopolysaccharide
Polyketides	PK	Tetracycline, aflatoxin B <sub>1</sub>