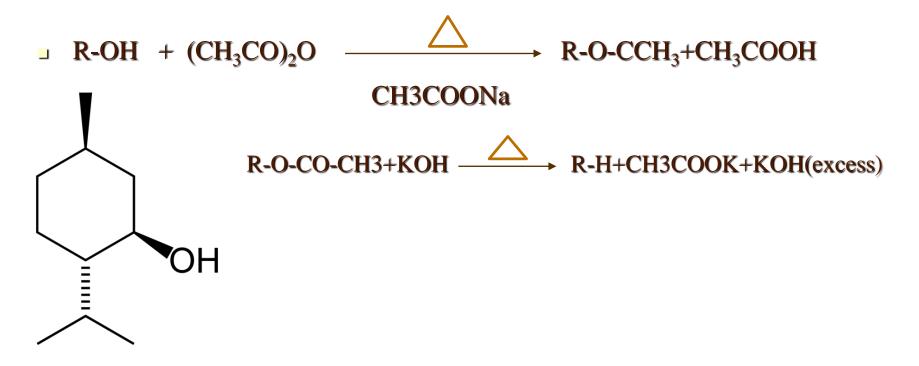
QUANTIFICATION OF TOTAL ALCOHOLS IN OLEUM MENTHAE

- Oleum Menthae is an essential oil which is obtained using the dried leaves and flowering tops of *Mentha piperita* L.
- The essential oil consists of % 50 % menthol and 10 % mentone. Menthyl acetate and menthofurane are of the other constituents of Oleum Menthae.

The essential oil is initially acetylated for the quantification of alcohols, thus all the alcohols turn into ester getirilir.



- In case the essential oil contains both free alcohols and esters, the ester value is determined by saponification prior to the acetylation.
- Following the acetylation, the ester value of the oil is calculated. This calculation yields the total of ester and esterificated alcohol.
- The difference of these two results is the free alcohol amount.
- Ester value: the number of milligrams of potassium hydroxide required to saponify the esters present in 1 g oil.

Experiment Procedure

10 ml oil + 1 g anhydr Na-acetate +10 ml acetic acid anhydride

1 hour reflux 3 kez shake with 5 ml hot water in a seperation funnel (3 times)

Lower phase (water phase) is discarted

Upper phase is washed with $Na_2CO_3 TS$ and 1:1 water: Na_2CO_3 in seperation and lower phases are combined. This process is repeated untill pink color occurs with 1 drop of phenolphthalein

The oil is washed with distilled water once and dried using anhydr Na_2SO_4

5 ml of the oil is weighed

+

50 ml 0.5 N KOH is added to 5 ml of the oil and refluxed for 30 minutes (saponification)

+

10 drops of phenolphthalein is added and excess of KOH is titrated with 0.5N HCl untill the pink color fades

• total alcohol % = $7,813 \times A \times (1-0,0021 \times E)$ (B-0,021 X A)

B= weight of the acetylated oil (5 ml)
E= percentage of ester calculated for menthyl acetate
A= b-a amount of 0.5 N HCl consumed for blank (a ml) amount of 0.5 N HCl consumed for the sample (b ml)

B-0,021 X A : An experimental value which converts the weight of acetylated oil to the weight of oil.

1-0,0021 X E : The value which excludes the error occurs because of the components except menthol (It provides the determination of the weight of free esters).