

POLYSACCHARIDES

▶ **STARCH**

▶ **DEXTRAN**

▶ **GUM**

▶ **MUCILAGE**

▶ **CELLULOSE**

▶ **PECTIN**

▶ **MUCOPOLYSACCHARIDES**

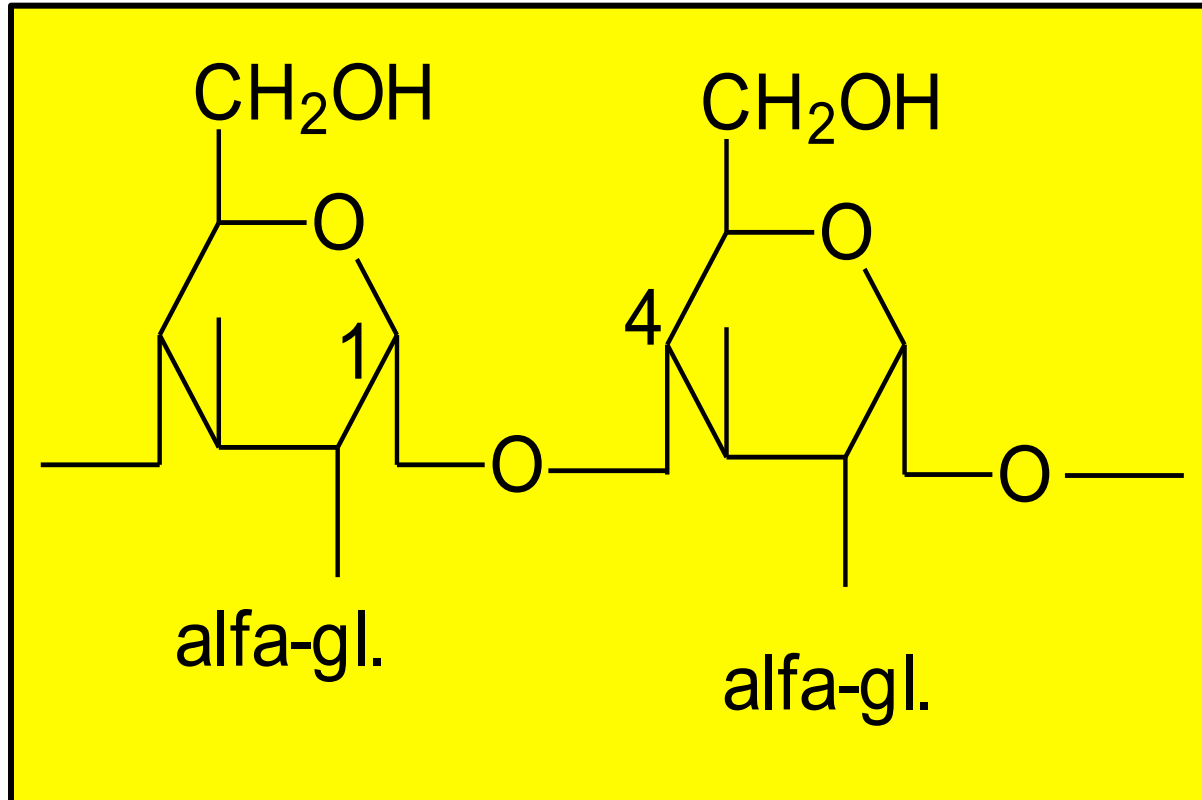
STARCHS

- ▶ **The main reserve substance in plants.**
- ▶ **These carbohydrates obtained from seeds, roots and fruits**
- ▶ **Tubers (Potato), stem (palm), roots (manihot) , rhizome (Maranta) , seed (Leguminosae and Graminae), fruit as reserve substance**
- ▶ **Especially found in seed and underground parts**

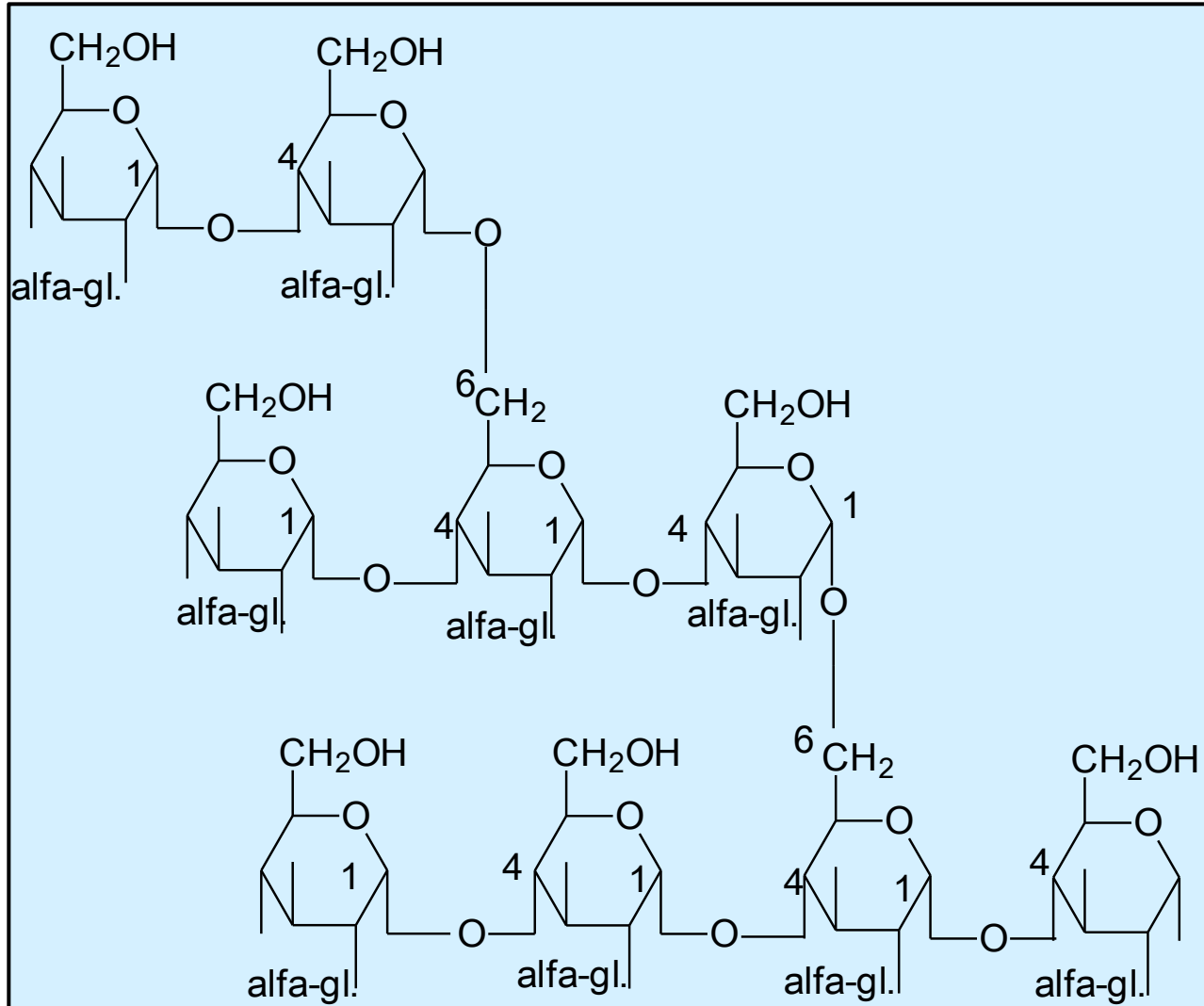
- ▶ **MW is in between 50.000-1.000.000**
- ▶ **Odourless, tasteless white powder**
- ▶ **Insoluble in cold water, dissolve partly in hot water and jellified**
- ▶ **Blue colour produce with iodine**
- ▶ **Composed almost pure glucose**
- ▶ **Hydrolysed with acid or enzymes resulted in glucose**

- ▶ The two main constituents are Amylose and amylopectin
- ▶ **Amylose**; soluble part of starch, %10-20 part of starch composed α -D glucose nonbranching helical chain bonded 1-4 glycosidic linkage
- ▶ Produce blue colour with iodine
- ▶ **Amylopectin**; insoluble in water, swelling part of starch, %80-90 of starch composed α -D glucose branched chain, 1-4 linkages in the chains and 1-6 linkages at the branch points
- ▶ Produce blue-violet colour with iodine

Amylose



Amylopectin



- ▶ 2 molecule of **α -D glucose** bonded 1-4 glycosidic linkage resulted in **MALTOSE** a disaccharide, occurs. And this molecules are the main structure of starch
- ▶ Three enzyme hydrolyzed the starch molecules to α -D glucose:
 - ▶ **1) α -Amylase**: Found in pancreas and saliva in animals. These enzyme induce to hydrolization to maltose and glucose molecules

NIŞASTALAR

- ▶ **2) β -amylase**: Induce hydrolization to maltose
- ▶ **3) α -1-6 Glucosidase**: Induce breaking up in brached structure of starch
- ▶ Starch.....diluted HCl....maltose..... α -glucose occurs

▶ Identification

- ▶ I_2 , Iodine-KI or iodine-water resulted in blue colour due to amylose structure. Amylopectin gives purple or brown colour with iodine
- ▶ **Soluble in %50 of chloral hydrate solution**
- ▶ “Blue value” for starch can be measured and used for identification
- ▶ Blue value : Absorbption of blue colour of iodine with starch at 680 nm. Blue value is proportional with amylose content. If amylose content is high blue value is high.

▶ HILUM

- ▶ The formation of starch granule starts at a point called the hilum in leucoplast
- ▶ Some of them are **centric** placed at the center of granule, some of them are **excentric** placed out of the center of the granules
- ▶ They can be so small and is not possible to seen

- ▶ **Starch granules come in a wide range of shapes, which assists identification , disc, spheres, ovals, elongated, rounded, kidney-shaped, polyhedral and irregular forms are common.**
- ▶ **They can be alone or combined**
- ▶ **Granules of starch jellyfied with water and gel time related to starch source and reaction conditions. These specifications can be determinant for identification of starch**
- ▶ **Gel temperature in water**
- ▶ **Gel time in %10 KOH solution is also important for determinant**

▶ Production of starch

- ▶ Drug is grained and mixed with water. Upper phase is transferred and lower phase is dried for obtaining starch and grained.
- ▶ Different sources are used according to the country and geographic places: Wheat, rice, potato, corn, *Maranta arundinacea* (Marantaceae), *Manihot esculenta* (Euphorbiaceae) (Tapioka starch and *Metroxylon* species (Palmae= Arecaceae) (Sago starch)

- ▶ ***Oryza sativa* (rice), *Zea mays* (corn) are the main sources for pharmacy**
- ▶ **Industry especially rice and wheat also corn are used (yield %65-75)**
- ▶ ***Solanum tuberosum* (potato) contains %20 of starch**

▶ Quantification

- ▶ **1) Gravimetric:** Obtained and weighed
- ▶ **2) Titrimetric:** Hydrolysed with acid or enzyme and reducing monosaccharides are quantified by titrimetric method
- ▶ **3) Colorimetric:** Starch+iodine, blue colour measured by colorimetry
- ▶ **4) Polarimetric:** Starch hydrolyzation and occurring monosaccharides are measured by polarimetry

▶ Specifications of Starch used in Pharmacy

- ▶ 1) Does not contain oil and impurity
- ▶ 2) Granules of starch are not deformed
- ▶ 3) Ash content should be lower than % 0.6
- ▶ 4) Ash content should be lower than %15
- ▶ 5) Nitrogen content should be defined
- ▶ 6) Acidity should be controlled
- ▶ 7) Microorganisms and fungi are not found

▶ USAGE

- ▶ 1) Humectant and main component of talc. Used against irritatant and refreshing activity on inflammation
- ▶ 2) Digestible easily therefore used as nutrient for infants and elder patients.
- ▶ 3) Small granule starchs is more suitable for talc such as rice

- ▶ **4) Diluent in powders and tablets manufacture**
- ▶ **5) Starting material for glucose, dextrin and amyl alcohol**
- ▶ **6) Antidote for iodine poisoning**
- ▶ **7) Hydroxy ethyl starch used as %6 solution in isotonic saline solution for treatment of hypovolemic shock**

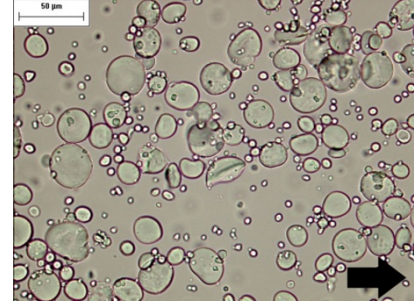
▶ PREPARATIONS

- ▶ 1) Fresubin diabetes (diet): %70 starch+%30 fructose
- ▶ 2) Nutrodrip diabetes (diet): Starch+fructose+ medium chain triglyceride
- ▶ 3) Undo talk

▶ **1) AMYLUM TRITICI (TF) (Wheat starch):**

▶ *Triticum vulgare* (*T. sativum*) (Graminae)

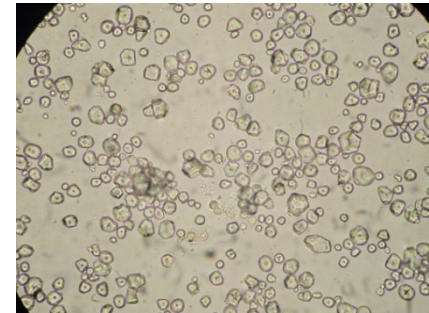
▶ Lentiform, 15-45 μm , hilum seen hardly.



▶ **2) AMYLUM MAYDIS (Corn starch):**

▶ *Zea mays* L. (Graminae)

▶ Polyangular granules, 10-25 μm , hilum is centric and cracked



▶ **3) AMYLUM SOLANI**

▶ **(Potato starch) (EP):**

▶ ***Solanum tuberosum* L. (Solanaceae)**

▶ Ovoid or pear shaped, 15-100 μm

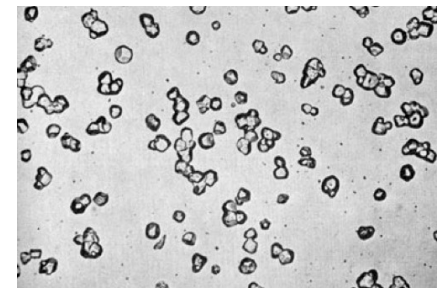
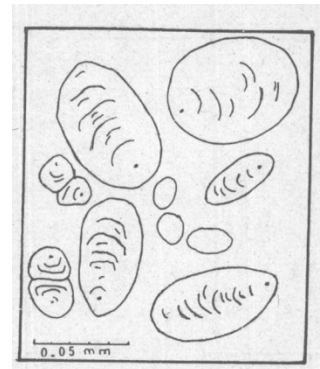
▶ Two or three granules are combined, hilum excentric

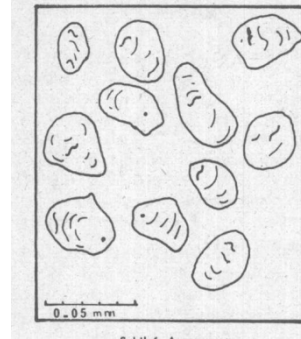
▶ **4) AMYLUM ORYZAE (TF) (Rice starch):**

▶ ***Oryza sativa* L. (Graminae)**

▶ Polyangular shaped, granules are simple or many of them combined, 2-10 μm ,

▶ hilum centric





▶ **5) ARROWROT (Hararot) (Amylum Maranthae):**

▶ *Maranta arudinacea* (Maranthaceae) rhizome

▶ Similar to potato starch, 30-40 μ m

▶ *Curcuma* sp. (Zingiberaceae) e.e.

▶ *Manihot utilissima* (Euphorbiaceae) tubers

▶ Arrowrot 60 $^{\circ}$ C soluble, 70 $^{\circ}$ C completely decomposed.

▶ Used as nutrient for infant food

▶ **AMYLUM SOLUBILE (Soluble Starch) (BP 1999):**

- ▶ Starch boiled in long time or
- ▶ Hydrolyzed with dil. AcidAmylopectin chain is decomposed and more soluble form in water is obtained

▶ **7) ULTRA AMYLOPECTIN:**

- ▶ Sodium amylopection gluconate
- ▶ Disintegrant in tablet manufacture
- ▶ Supplement material for washable pomade and pasta preparation
- ▶ Lubricant in pharmaceutical technology

- ▶ **8) AMYLUM NON-MUCILAGINOSUM (ANM):**
Tetramethylacetylene ether of starch
- ▶ Excipient for talc, powders and tablets
- ▶ **9) FARINA:** Seed flours which contain starch
- ▶ Used in pharmaceutical industry.

DEXTRIN

- ▶ **Products obtained by partial hydrolysis of starch**
- ▶ **Hydrolization products are; soluble starch, amyloextrine, achrodextrin and maltose**
- ▶ **Sweet taste, white/yellow, odourless, amorphous powder**

- ▶ Adhesive in pharmaceutical technology (Dissolved in 3 part of boiled water)
 - ▶ Agent for emulsion and suspension
 - ▶ Manufacture of tablets and capsule
 - ▶ Costmetics
 - ▶ Source for saccharides as nutrient.
- Does not contain lactose,
Contain low electrolyte
- ▶ Paper, ink, match, fabric and paint industry.

DEXTRIN

- ▶ *Oryza sativa glutinosa* and *Sorghum vulgare glutinosa* used as sources to obtain
- ▶ Gives red colour with iodine (Difference between dextrin and amylopectin)
- ▶ Starch...dil.acid (HNO_3)....**DEXTRIN** (Dried at 110-120 °C White colour)
- ▶ (Dried at 150-250 °C Yellow colour) (More hydrolyzed, contains more maltose molecules)

CYCLODEXTRINS

- ▶ **Cyclic oligosaccharides produced by the enzymatic degradation of starch (generally 6-8).**
- ▶ **The enzyme cyclodextrin glycosyl transferase is produced by different *Bacillus* sp.**
- ▶ **Degradation by this enzyme is produced**
- ▶ **Cyclodextrins are cyclic glucose units and shaped like a torus**

CYCLODEXTRINS

Structurally α -, β -, and γ - cyclodextrins consist of six, seven and eight glucose units respectively linked by α 1-4 bond

α - cyclodextrin (α CD= C 6 A)

▶ Cyclohexaglucans

▶ Cyclic structure consisting 6 α -glucopyranose units

β - cyclodextrin (β CD= C 7 A) (BP/USP/EP)

▶ Cycloheptaglucans

▶ Cyclic structure consisting 7 α -glucopyranose units

γ - cyclodextrin (γ CD= C 8 A)

▶ Cycloooktaglucans

▶ Cyclic structure consisting 8 α -glucopyranose units

- ▶ Molecular weight and solubility of cyclodextrins in water are different
- ▶ β - cyclodextrin (β CD) the least dissolved in water
- ▶ γ - cyclodextrin (γ CD) the most dissolved in water

USAGE

▶ **Pharmaceutical technology**

- ▶ **1) Excipient (diluent, to increase solubility, tablet manufacturing)**
- ▶ **2) Formulation in oral dosage forms:**
 - ▶ **a) To prepare crystalline form of liquid compounds**
 - ▶ **b) To mask taste and smell Kötü koku, tat düzeltici**
 - ▶ **c) Producing complex of compounds to improve their miscibility.**

- ▶ **3) Improving physical and chemical specifications:**
- ▶ **a) Increasing stability of volatile compounds**
- ▶ **b) Protect compounds against oxidation**
- ▶ **c) Prevent decomposition, polymerisation and catalytic reactions**
- ▶ **d) Desensitization against light, acidity of stomach and others**

- ▶ **4) Improve bioavailability:**
- ▶ **a) Improve solubility in water**
- ▶ **b) Modify serum concentration of hydrophobic compounds as their complex with cyclodextrin after oral administration to decrease their required dosage**
- ▶ **c) Decrease hydrophobicity by producing complex. Increase percutaneous and rectal absorption**
- ▶ **d) Prepare of liquid drug formulations (parantral solutions (injectable), eye drops and others)**

▶ **Food Industry**

- ▶ **1) Stabilization of taste**
- ▶ **2) Protect microbial contamination**
- ▶ **3) Mask taste and smell**
- ▶ **4) Prevent oxidation, thermal and light decomposition, sublimation and volatility**

DEXTRANS

- ▶ Polysaccharides consisting α -D-glucopyranose molecules occurred by proliferation of some microorganisms such as *Leuconostoc mesenteroides* in medium containing sucrose.
- ▶ For usage in pharmacy;
- ▶ 1) *Leuconostoc mesenteroides* NRRLB 512 strain must be used
- ▶ 2) α -glucose molecules must be 1-6 bonded
- ▶ 3) MW should be 100×10^6 olmalı

▶ Production of dextran

- ▶ Two different methods are used in industry:
- ▶ 1. Method: %10 sucrose containing sterile medium is used and left for 24 h. **DEXTRAN** is obtained by %50 yielding Ethyl alcohol+CH₃COOH+Lactic acid obtained as by-products
- ▶ Methyl alcohol is added to precipitate **DEXTRAN**
- ▶ **DEXTRAN** + non pyrogen water dissolved and then precipitate again

- ▶ Partial hydrolysis resulted in desired MW of DEXTRAN as 40.000 and 75.000
- ▶ After precipitation of methyl alcohol and recrystallization medicinal dextran is obtained
- ▶ **2. Method: Dextran sucrose is an enzyme and production of *Leuconostoc mesenteroides*. *In vitro* dextran sucrose catalyses sucrose solution to production of dextran**

%2 sucrose + %2 corn germ + mineral compounds containing medium used for fermentation with microorganism at pH=6.7 for 12 h

- ▶ **Dextran sucrose enzyme concentration reach to the maximum level, medium will filtered to remove microorganisms**
- ▶ **pH is adjusted to 5.0 and the temperature is 15 °C using dextran sucrose enzyme dextran is obtained from sucrose**
- ▶ **Sucrose.....dextran sucrose (*Leuconostoc mesenteroides*)...Glucose+Fructose polymerisationhigher molecular weight of **DEXTRAN**.....partial hydrolysis.....**DEXTRAN****

▶ USAGE

- ▶ **1) Two different type of dextran and their complex and ester forms are used**
- ▶ **2) Dextrans are administered intravenously (infusion) . The viscosity and osmolarity of these solutions are close to plasma**
- ▶ **Dextran is nontoxic, neutral serologically , prolonged action and completely eliminated. It is a plasma substitute used for following indications:**
- ▶ **For plasma volume expansion in due to hemorrhage, trauma, toxiiinfection**

- ▶ 3) Used in some preparations as stabilizer
- ▶ 4) In synthetic tears preparations against xerophthalmia
- ▶ (Dextran-70...Tears naturale;
Dextran-40.....Clarex drops)

- ▶ **Dextran production in desired molecular weight**
- ▶ **1) Diluted acid hydrolysis for partial depolymerisation**
- ▶ **2) Modulated precursor compounds are added (maltose or low molecular weight dextrans)**

- ▶ **Different dextrans**
- ▶ **Obtained by *Acetobacter*, *Streptobacterium*, *Lactobacillus*, *Streptococcus* strains**
- ▶ **1-6, 1-4 and 1-3 bonded. All are glucans or glucose polymers**
- ▶ **Molecular weight / degree of polymerisation / viscosity / solubility in water / monosaccharide bonds / polymerisation degree are the differences of dextrans**

DEXTRAN 40, DEKSTRAN 40, GENTRAN 40, RHEOMACRODEX

- ▶ **MW approximately 40.000 α 1-6 bonded dextran**
- ▶ **In surgery first application of the cardiopulmonary bypass pump**
- ▶ **Isotonic solution is used for preventing of agregation of erythrocyte aggregation**
- ▶ **Regulation of blood stream by decreasing viscosity**
- ▶ **%5 dextrosum solution containing %10 Dextran 40 solution is used in hypovolemic shock**

DEXTRAN 40, DEKSTRAN 40, GENTRAN 40, RHEOMACRODEX

- ▶ **It is also indicated that for dehydration and extensive burns**
- ▶ **Side effects;**
- ▶ **Hypersensitivity reactions are rare but always possible, thus the infusion must begin very slowly. To prevent or alleviate the anaphylactic reaction first it is preferable to inject a very low molecular weight dextran which block the antigen sites on antibody.**
- ▶ **for patients that have kidney problems, pulmonary edema is not suitable**

DEXTRAN 75, DEKSTRAN 75, GENTRAN 75, MACRODEX

- ▶ **MW approximately 75.000, α 1-6 bonded dextran**
- ▶ **% 6 dextran solution in isotonic saline or %5 glucose(dextrose) is used for plasma volume expansion in surgery and extensive wounds. However if the blood cells are decreasing or in case of lipoproteinemia dextran solution usage is not suitable**
- ▶ **Hypovolemic shock**
- ▶ **Clinically anticoagulant activity is important. Used against trombosis**

Semi-synthetic derivatives of dextran

▶ DEXTRAN SULPHATE

- ▶ H_2SO_4 esther of dextran
- ▶ Anticoagulant in trombosis and phlebitis as pomade externally

▶ IRON-DEXTRAN ENJECTABLE SOLUTION

- ▶ Small MW of dextran (partial hydrolyses of dextran) and $\text{Fe}(\text{OH})_3$ colloidal and sterile complex solution
- ▶ **Production:** Dextran water solution+ FeCl_3 addition.....mixture cooling....centrifuged.....iron content adjusted to %95-105.....concentrated.....filtered and sterilized
- ▶ Dark brown colour.
- ▶ Used for anaemia
- ▶ IM form is easily absorb without irritation



IRON SORBITOL (=JECTOFER)

- ▶ **Sterile solution**
- ▶ **Fe+Sorbitol+citric acid complex**
- ▶ **Used for treatment of anaemia**