## Practice 32.11.

Ephedrine Hydrochloride Nasal Drop (Remington $20^{\text {th }}$ ed. )

| Ephedrine Hydrochloride | 0.5 g |
| :--- | :---: |
| Chlorobutanol | 0.5 g |
| Sodium chloride | 0.5 g |
| Purified water | q.s. |

## Practice 32.12.

Oxymetazoline Hydrochloride Nasal Spray

| Oxymetazoline Hydrochloride | 5 mg |
| :--- | :--- |
| Benzalkonium chloride | 1.5 g |
| EDTA | 5 mg |
| pH 6.0 phosphate buffer | 10 ml |

## Pratice 32.4.

Silver Nitrate Eye Drop
15 ml , prepare an isotonic $1 \%$ silver nitrate solution.

## Practice 32.15.

Aluminum Sulphate Ear Drop

| Aluminum Sulphate | 225 g |
| :--- | :---: |
| Acetic acid $(\% 33)$ | 250 ml |
| Tartaric acid | 45 g |
| Calcium carbonate | 100 g |
| Purified water | 750 ml |

## Practice 32.16.

Phenolic Ear Drop

| Phenol-Glycerine* |  | 40 ml |
| :--- | ---: | ---: |
| Glycerine | q.s. | 100 ml |


| * Phenol-Glycerine |  |
| :--- | ---: |
| Phenol | 160 g |
| Glycerine | 840 g |

## Practice 32.17.

Sodium Bicarbonate Ear Drop (Remington $20^{\text {th }}$ ed.)
Sodium bicarbonate 5 g
Glycerine
30 ml
Purified water q.s.
100 ml

## Practice 31.10.

Dialysis Solution

## Solution I

Sodium chloride
Potassium chloride
Calcium chloride dihydrate
Magnesium chloride hexahydrate
Acetic acid (\%100)
Water obtained by reverse osmosis method

## Molecular weight

### 214.800 g <br> 58.5

$2.612 \mathrm{~g} \quad 74.6$
7.720 g
147.0
203.0
3.558 g
60.0
q.s. $\quad 1000.000 \mathrm{ml}$

## Solution II

Sodium bicarbonate
84.0 g
84.0
q.s. $\quad 1000.0 \mathrm{ml}$

## Preparation:

Solution I and Solution II are mixed at specific ratios and diluted with water obtained by reverse osmosis. (Solution I + Solution II + Water obtained by reverse osmosis method: 1 liter +1.225 liter +32.775 liters)

## Questions:

1. Why should this formulation be prepared with water obtained by reverse osmosis?
2. Calculate the amounts of $\mathrm{mEq} / 1, \mathrm{mmol} / 1$ and total solution osmolarities of $\mathrm{Na}^{+}, \mathrm{K}^{+}, \mathrm{Ca}^{+2}, \mathrm{Mg}^{+2}$, $\mathrm{HCO}_{3}^{-}, \mathrm{Cl}^{-}$, and $\mathrm{CH} 3 \mathrm{COO}^{-}$in the prepared solution.
3. What is the pH of the diluted solution?
4. How is this solution used?
5. What are the characteristics of the ideal dialysis solution?
