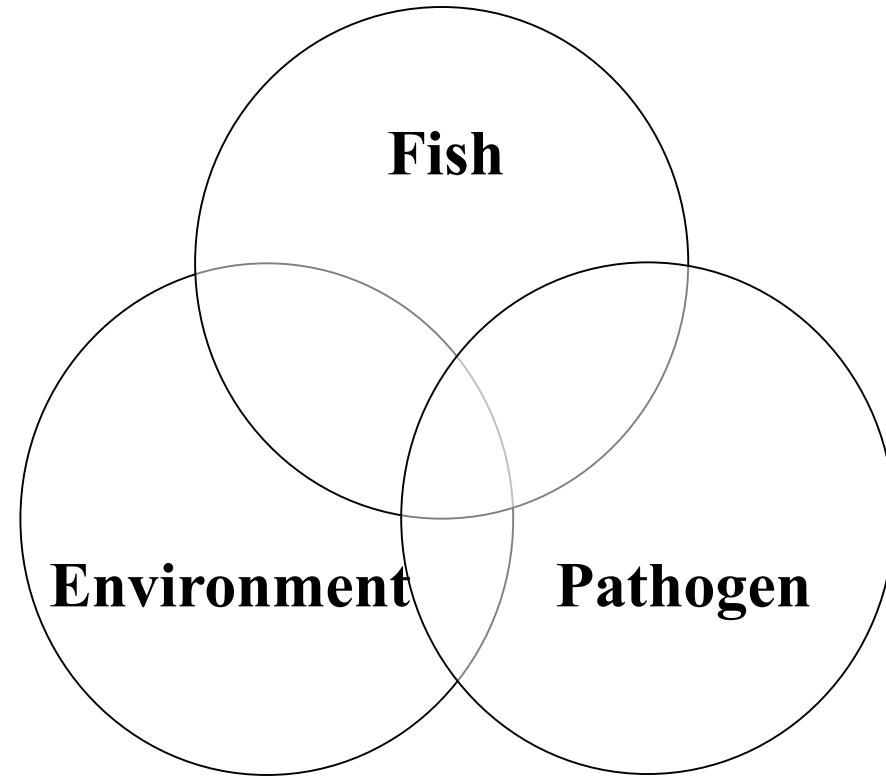


## **2. The water quality in aquaculture and the selection of culture site**

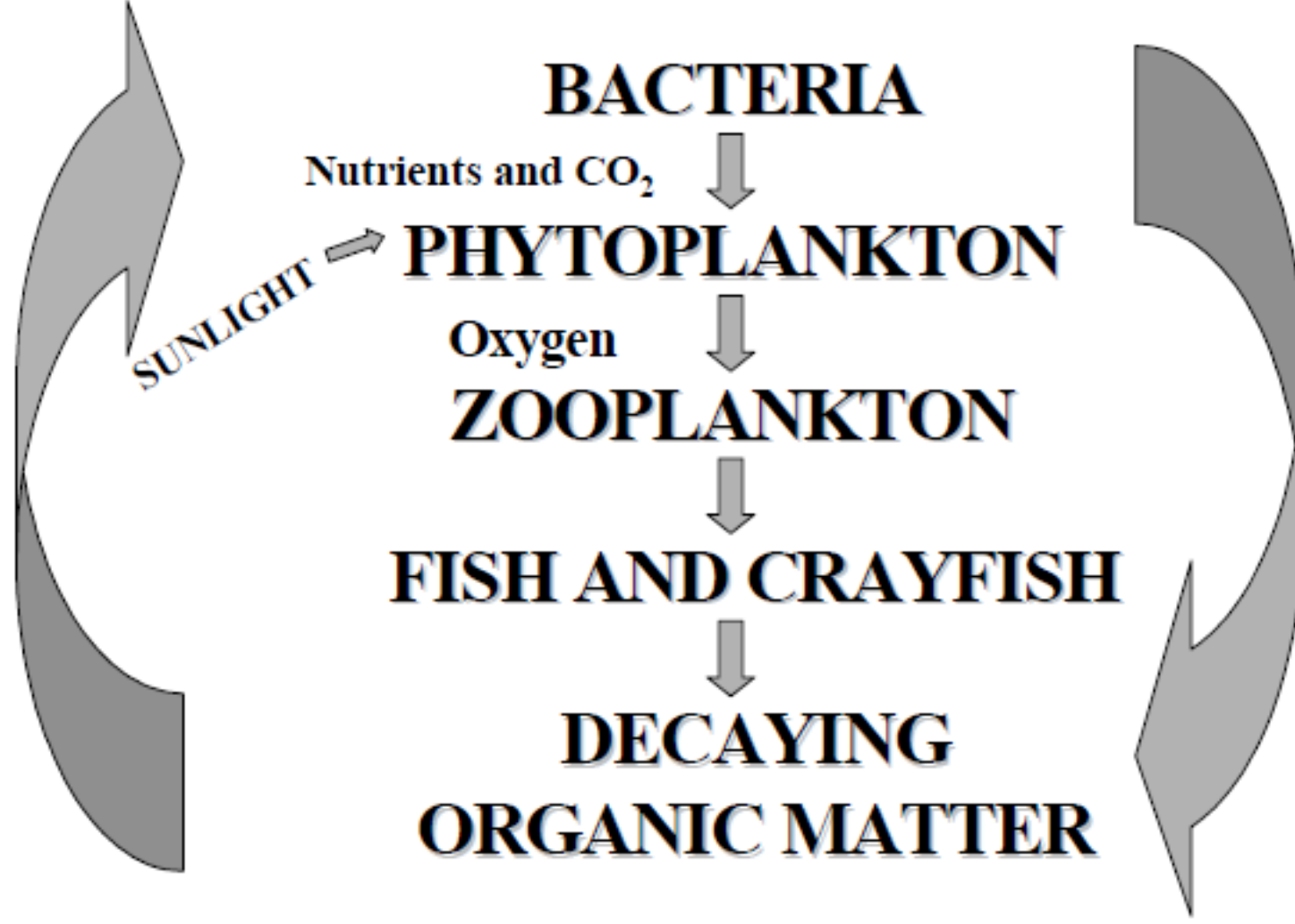
# Fundamentals for optimal performance and on-farm disease prevention and control

- Good husbandry
- Good nutrition
- Good genetic stock
- Good management
- Good environment
- Good bio-security



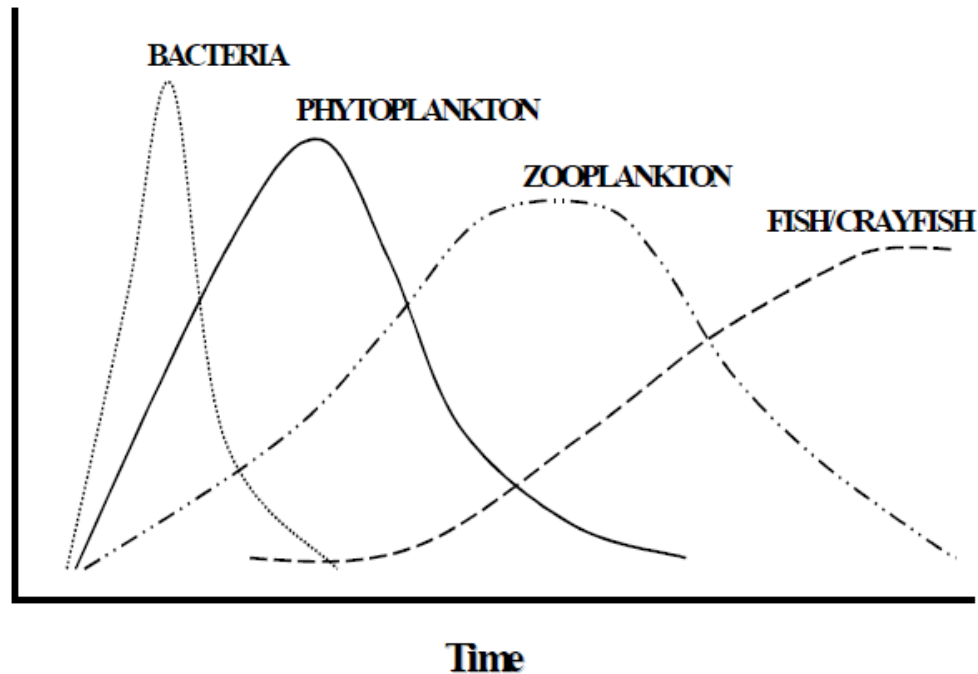
Taken From:

<http://images.pcmac.org/SiSFiles/Schools/AL/MobileCounty/BryantHigh/Uploads/Presentations/WATER%20QUALITY%20AND%20MANAGEMENT.ppt>

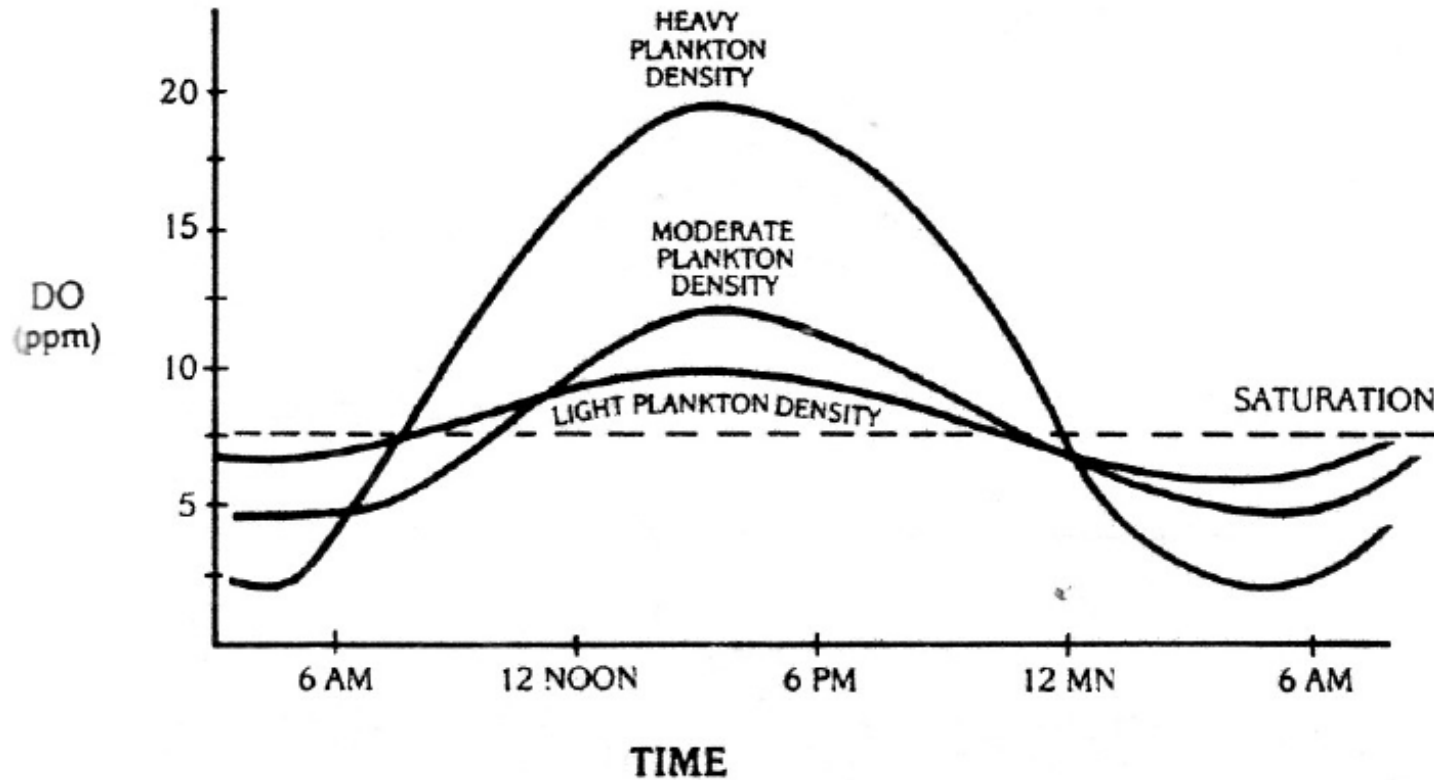


# Succession

- The aquatic organisms within an aquaculture pond will vary over time
- It is therefore important to have a good understanding of the population dynamics within your pond to stabilise population numbers of aquatic organisms and to ensure that the system will not crash.



# Changes in Oxygen level due to different algae density

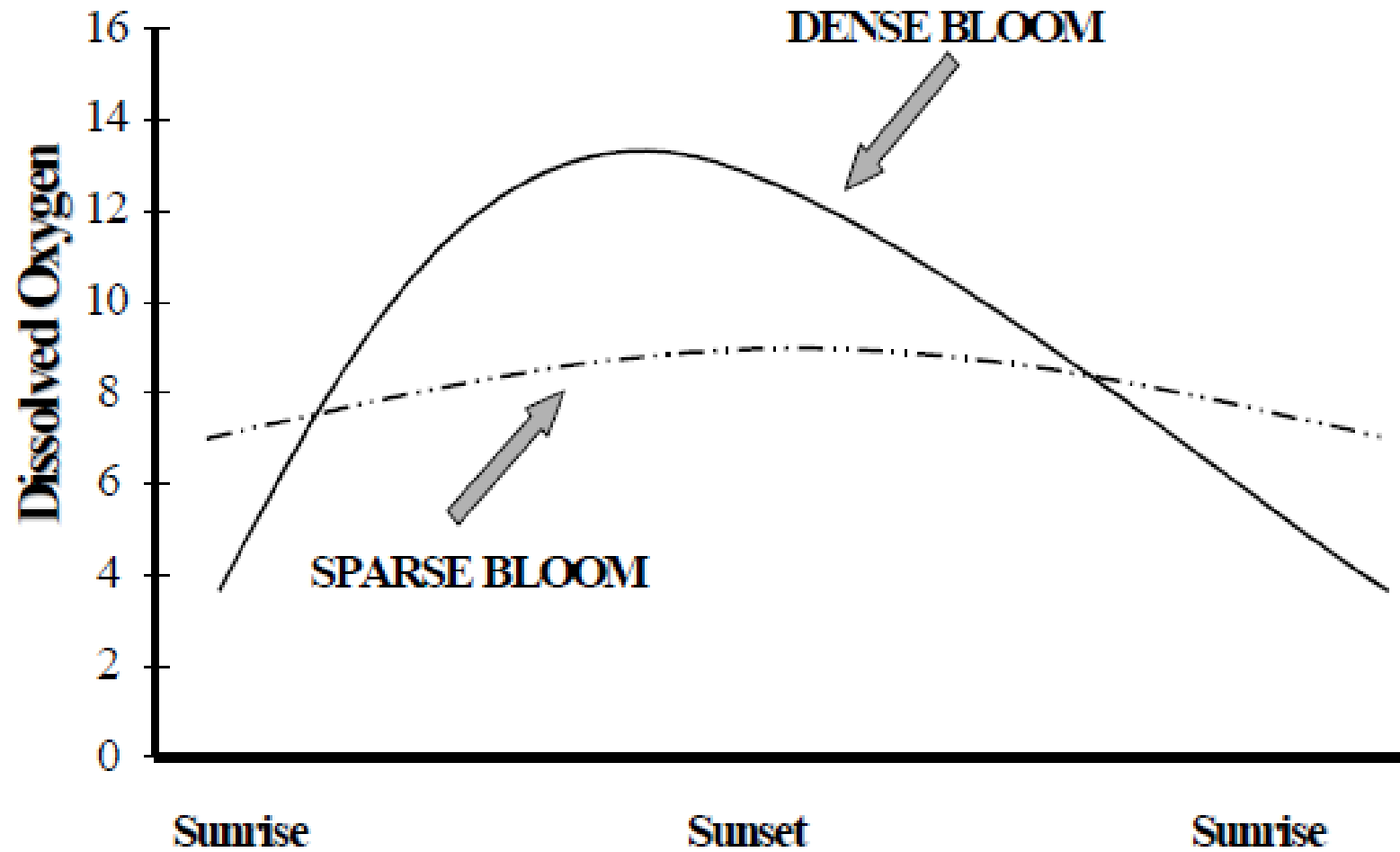


Taken From:

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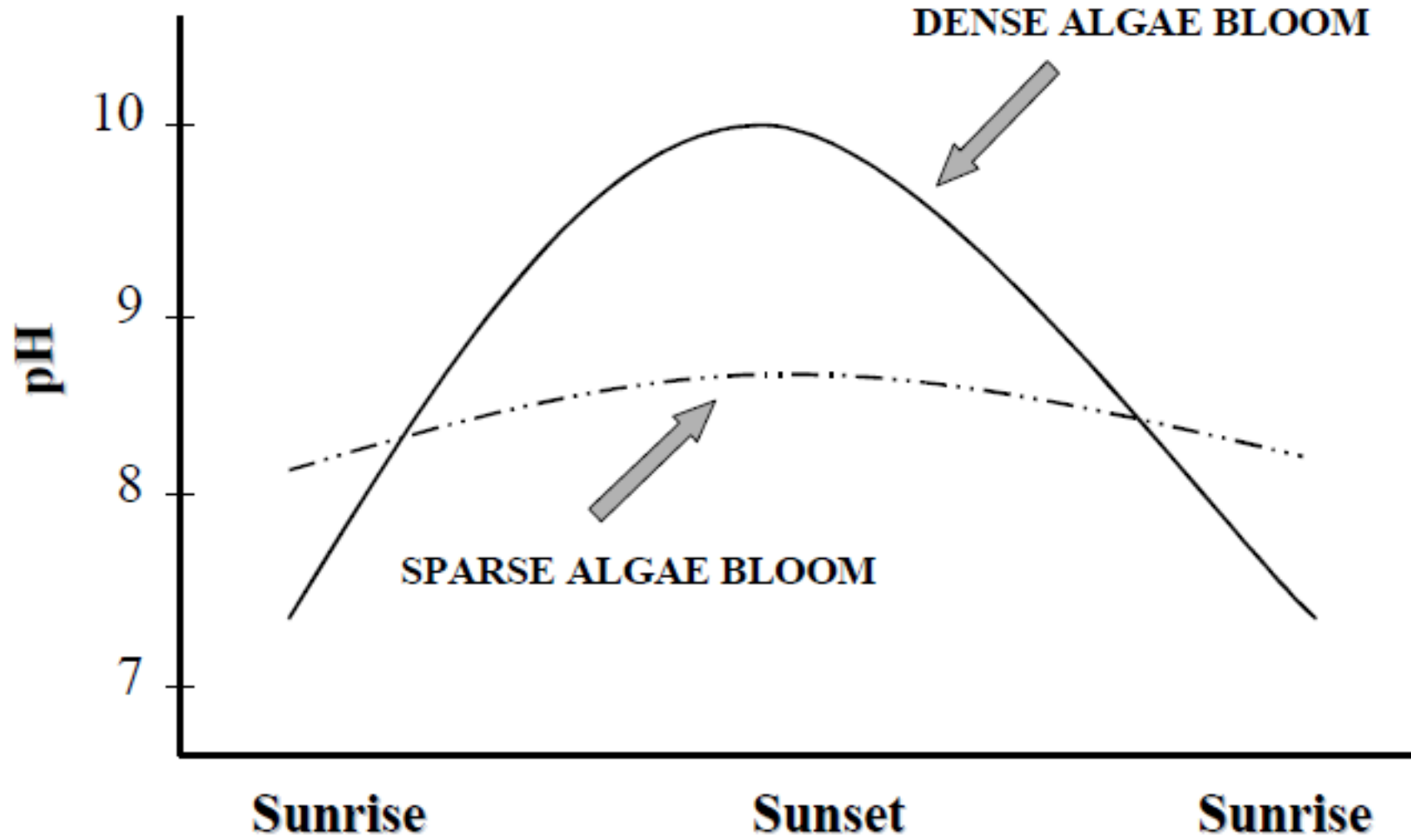


# Oxygen dynamics



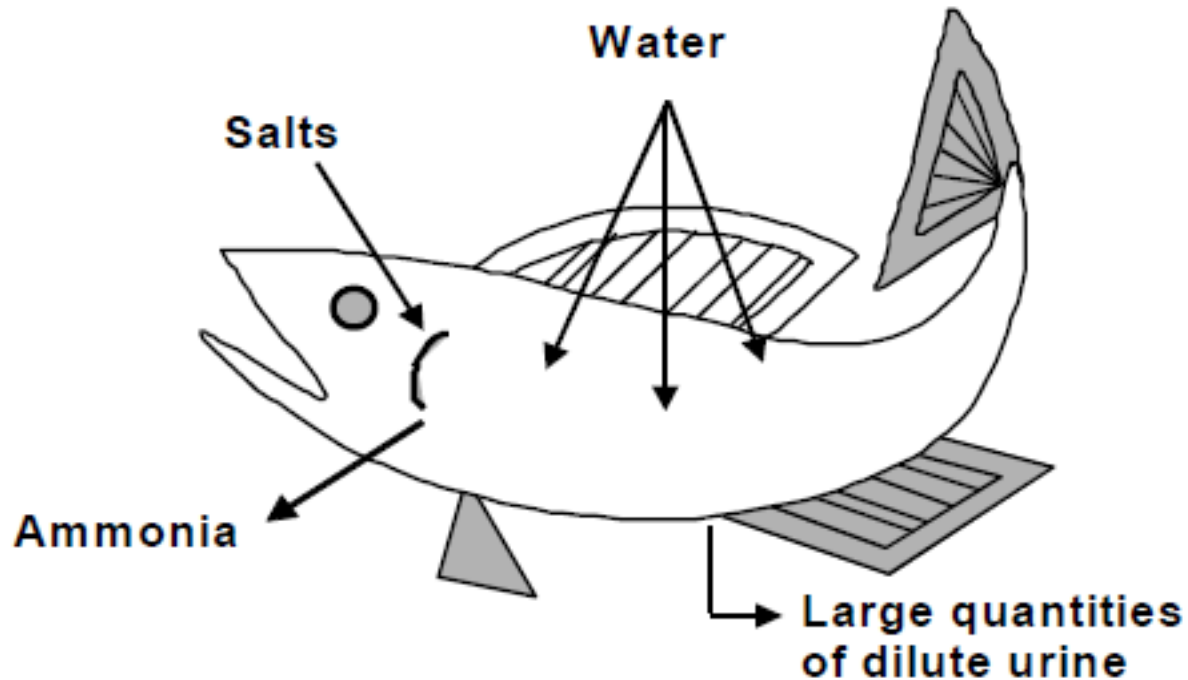


# pH daily fluctuations



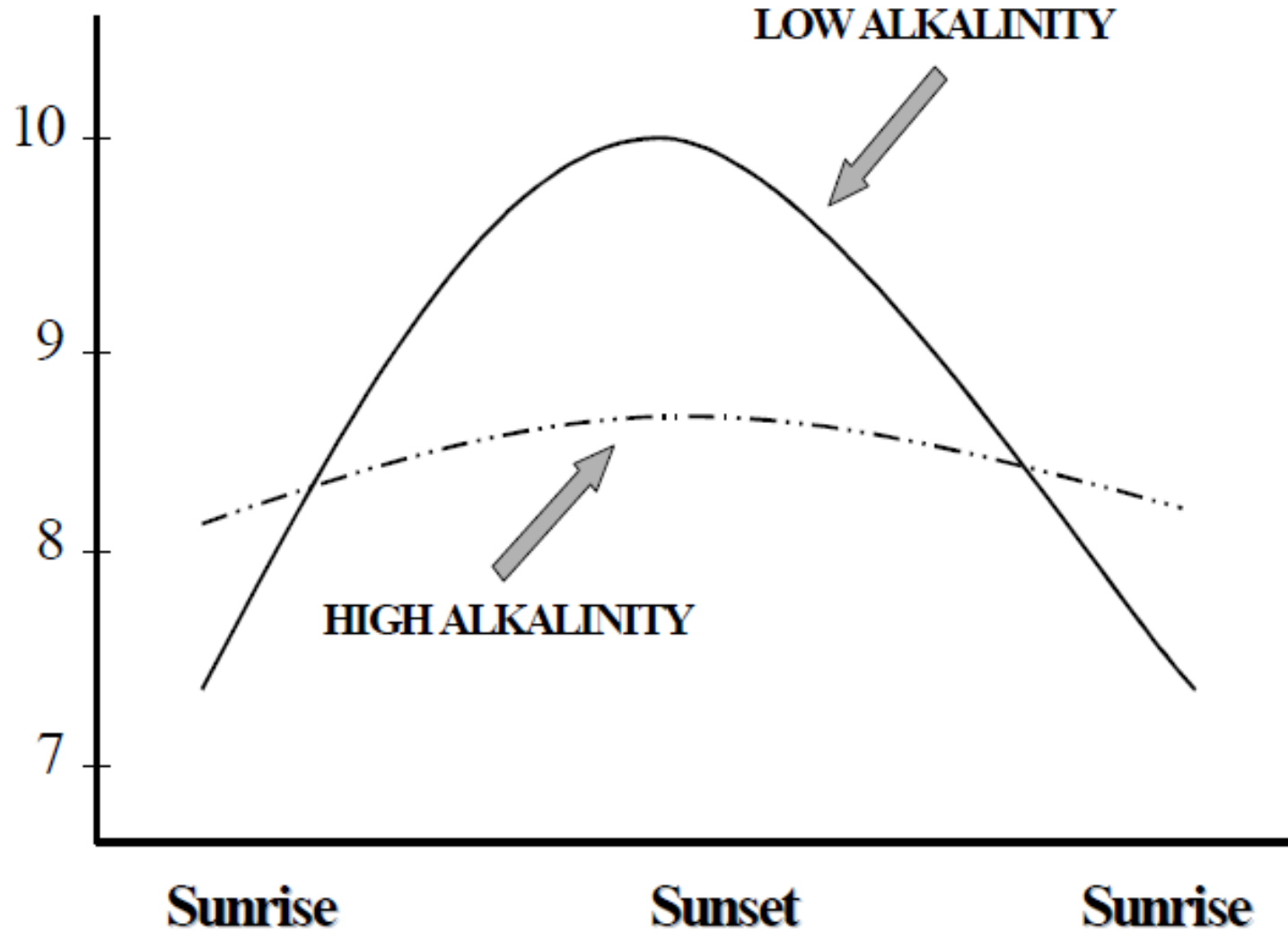
# Osmoregulation

- A freshwater fish will gain water via osmosis.
- Excess water is excreted in the urine and ion uptake is through the gills
- If salinity is too high, the fish will start to lose water to the environment. As freshwater fish are not physiologically adapted to osmoregulate within a saline water source, decreased growth and survival can occur under these conditions





# Alkalinity changes in different types of pond



Taken From:

<http://images.pcmac.org/SiSFiles/Schools/AL/MobileCounty/BryantHigh/Uploads/Presentations/WATER%20QUALITY%20AND%20MANAGEMENT.ppt>

# Water quality limits

Element	Form in Water	Desired Concentration
Oxygen	Molecular Oxygen (O <sub>2</sub> )	5 – 15 mg/l
Hydrogen	H <sup>+</sup> [-log(H <sup>+</sup> ) = pH]	PH 7 – 9
Nitrogen	Molecular Nitrogen (N <sub>2</sub> ) Ammonium (NH <sub>4</sub> <sup>+</sup> ) Ammonia (NH <sub>3</sub> ) Nitrate (NO <sub>3</sub> <sup>-</sup> ) Nitrite (NO <sub>2</sub> <sup>-</sup> )	Saturation or less 0.2 – 2 mg/l < 0.1 mg/l 0.2 – 10 mg/l < 0.3 mg/l
Sulfur	Hydrogen Sulfide (H <sub>2</sub> S) - rotten egg gas Sulfate (SO <sub>4</sub> <sup>-</sup> )	Not detectable 5 – 100mg/l
Carbon	Carbon Dioxide (CO <sub>2</sub> )	1 – 10 mg/l
Calcium	Calcium Ion (Ca <sup>2+</sup> )	5 – 100 mg/l Can be higher in crustacean ponds
Magnesium	Magnesium ion (Mg <sup>2+</sup> )	5 – 100 mg/l
Sodium	Sodium ion (Na <sup>+</sup> )	2 – 100 mg/l
Potassium	Potassium ion (K <sup>+</sup> )	1 – 10 mg/l
Bicarbonate	Bicarbonate ion (HCO <sub>3</sub> <sup>-</sup> )	50 – 300 mg/l

Boyd (1998) "Water Quality for Pond Aquaculture"

# Water quality limits

Element	Form in Water	Desired Concentration
<b>Carbonate</b>	Carbonate ion ( $\text{CO}_3^{2-}$ )	0 – 20 mg/l
<b>Chloride</b>	Chloride ion ( $\text{Cl}^-$ )	1 – 100 mg/l
<b>Phosphorus</b>	Phosphate ion ( $\text{HPO}_4^{2-}$ , $\text{H}_2\text{PO}_4^-$ )	0.005 – 0.2 mg/l
<b>Silicon</b>	Silicate ion ( $\text{H}_2\text{SiO}_3$ , $\text{HSiO}_3^-$ )	2 – 20 mg/l
<b>Iron</b>	Ferrous iron ( $\text{Fe}^{2+}$ ) Ferric iron ( $\text{Fe}^{3+}$ ) Total iron	0 mg/l Trace 0.05 – 0.5 mg/l
<b>Manganese</b>	Manganese ion ( $\text{Mn}^{2+}$ ) Manganese dioxide ( $\text{MnO}_2$ ) Total manganese	0 mg/l Trace 0.05 – 0.2 mg/l
<b>Zinc</b>	Zinc ion ( $\text{Zn}^{2+}$ ) Total zinc	< 0.01 mg/l 0.01 – 0.05 mg/l
<b>Copper</b>	Copper ion ( $\text{Cu}^{2+}$ ) Total copper	< 0.005 mg/l 0.005 – 0.01 mg/l
<b>Boron</b>	Borate ( $\text{H}_3\text{BO}_3$ , $\text{H}_2\text{BO}_3^-$ )	0.05 – 1 mg/l

**Boyd (1998) “Water Quality for Pond Aquaculture”**