## Experiment -1 Hydrostatic Bench

The Hydrostatic Bench enables the study of the main properties and the behavior of such liquids under hydrostatic conditions, with the aid of some accessories to make the different experiments.

## Equipment Description

The equipment consists of a metallic structure assembled on wheels with a panel at the top. In the lower part of the bench there is a tank where water is stored. Water is then sent to a methacrylate tank placed at the upper part of the bench and to other plastic deposit. Two hand-operated pumps are used for such distribution. The methacrylate tank is connected to two communicating tubes on the front panel, enabling to perform some practices; the other deposit placed on the horizontal surface of the bench is necessary for performing the rest of the practices. All water in excess is sent back to the storage tank by the drain. The rest of the equipment consists of the following different elements and independent accessories:

- Barometer (10)
- Thermometer (3)
- Ubbelohde capillary viscosimeter, $0.6-3 \mathrm{cp}(0 \mathrm{c})$
- Ubbelohde capillary viscosimeter, 2-10 cp (I)
- Ubbelohde capillary viscosimeter, 10-50 cp (Ia)
- Ubbelohde capillary viscosimeter, 60-300 cp (IIc)
- 3 graduated cylinders
- Accessory for demonstration of free surface in static conditions (7)
- Bourdon manometers calibration (13)
- Mercury manometers (9)
- Accessory to determine the metacentric height (FME11)
- Accessory for studying Archimedes' principle
- Accessory for studying the hydrostatic pressure (FME08) (14)
- Fluid level gauge calibrator (16)
- Set of weights ( $5,10,20,50,100,400,1000,2000,5000$ gr.)
- Air pump
- 2 water pumps (11 and 12)
- Universal hydrometer (1)
- Chronometer
- Set of measurement cylinders (2 of 600 ml ) (4)
- Spare parts for the viscosimeter elements


Figure 2.0.1. Main parts of the Hydraulic Bench


## Experiment - 1.1

## Density and Specific Gravity Measurements

## Aim of this Experiment

To determine density and specific gravity.

## Necessary devices

Universal hydrometer.
Open precipitate tubes or cylinder


## Procedure

1. Fill the precipitate tube or cylinder with water in such a way that the hydrometer floats. Check that the submerged length corresponds to 1.00 in the graduated scale.
2. Fill the other three cylinders with the liquids to work with, and note down the scale mark for each one. This value in the scale indicates the specific gravity.
3. Note down the results obtained in the following graph, taking into account the values of the atmospheric pressure and temperature in the moment of performing the practice.

Pressure $\qquad$ mm Hg
Temperature $\qquad$ ${ }^{\circ} \mathrm{C}$

## Sample Test Results

| Liquid | Specific gravity | Density |
| :---: | :---: | :---: |
| Water |  |  |
| Glycerine |  |  |
| Motor oil |  |  |
| Oil |  |  |

## Experiment - 1.2

## Viscosity measurement

## Aim of this Experiment

Determine the viscosity of different liquids at atmospheric pressure and environmental temperature.

## Necessary devices

- Ubbelohde capillary viscosimeter, $0.6-3 \mathrm{cp}$
- Ubbelohde capillary viscosimeter, 2-10 cp
- Ubbelohde capillary viscosimeter, $10-50 \mathrm{cp}$
- Ubbelohde capillary viscosimeter, 60-300 cp
- Chronometer
- Hydrometer
- Thermometer



## Procedure

The liquids to be studied are:

- Car Motor oil
- Glycerol
- Castor oil

1. Find in tables four liquids of known viscosity, each one inside the measurement range of each viscosimeter.
2. Fill each Ubbelhode capillary viscosimeter, with the same volume of liquid of known viscosity and density, and write down the time used by the liquid of going down the viscosimeter.
3. Make four problem samples aliquots, with the same volume as used with known viscosity solutions. Measure their falling time in each viscosimeter. In some cases the liquid will fall too fast to take any measurement, and in others it will probably spend too much time. Avoid these liquids.
4. Write down the existing atmospheric pressure and temperature in that moment in the laboratory. With the aid of the data and expressions given hereafter, complete the following table:

Barometric Pressure $\qquad$ mm Hg

Temperature $\qquad$ ${ }^{0} \mathrm{C}$

Car Motor oil density (depending on the manufacturer) $\qquad$ $\mathrm{g} / \mathrm{cm}^{3}$

Glycol density $125 \mathrm{~g} / \mathrm{cm}^{3}$
5. Write down in next table the values obtained with solutions of known viscosity and density.

| Liquid | Viscosity $\left(\mathrm{cP}=10^{-}\right.$ <br> $2 \mathrm{~g} / \mathrm{cm} \cdot \mathrm{s})$ | Density $\left(\mathrm{g} / \mathrm{cm}^{3}\right)$ | Time (s) |
| :---: | :---: | :---: | :---: |
| A (range $0.6-3 \mathrm{cp}$ ) |  |  |  |
| B (range 2-10 cp ) |  |  |  |
| C (range $10-50 \mathrm{cp}$ ) |  |  |  |
| D (range 60-300 cp ) |  |  |  |

6. Now, repeat again the experience, with the problem samples, and fill next table with the obtained values, using previous obtained data and equations.

| Liquid | Time (s) | Density $\left(\mathrm{g} / \mathrm{cm}^{3}\right)$ | Viscosity $\left(\mathrm{cP}=10^{-2} \mathrm{~g} / \mathrm{cm} \cdot \mathrm{s}\right)$ |
| :--- | :--- | :--- | :--- |
| Motor car oill |  |  |  |
| Glycerol |  |  |  |
| Castor oil |  |  |  |

## Experiment -1.3

## Capillarity effect observation

## Aim of this Experiment

To observe the effect of the space between two plane surfaces with a capillary raising.

## Necessary devices

- Parallel Plates Capillary Device.



## Procedure

1. Clean carefully both glasses.
2. Loosen slightly the screws and vertically place strips between the glasses (These can be just pieces of paper).
3. Tighten carefully the screws.
4. Place the two glasses in the support guides.
5. Submerge in water.
6. Observe that where the space is smaller the raising is higher, and where the space is wider the raising is lower.
7. Do the same thing with other strips of different thickness.

## Appendix - I Useful Data

Table 1. Table of the atmospheric pressure in function of the height

| HEIGHT <br> $(\mathbf{m})$ | LEVEL OF <br> THE <br> VARIABLE | HEIGHT <br> $(\mathbf{m})$ | LEVEL OF <br> THE <br> VARIABLE | HEIGHT <br> $(\mathbf{m})$ | LEVEL OF <br> THE <br> VARIABLE |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 0 | 760 | 680 | 700.8 | 1360 | 645.2 |
| 20 | 758.2 | 700 | 698.9 | 1380 | 643.6 |
| 40 | 756.4 | 720 | 697.3 | 1400 | 642 |
| 60 | 754.6 | 740 | 695.5 | 1420 | 640.4 |
| 80 | 752.9 | 760 | 693.9 | 1440 | 638.8 |
| 100 | 751 | 780 | 692.4 | 1460 | 637.2 |
| 120 | 749.2 | 800 | 690.7 | 1480 | 635.6 |
| 140 | 747.4 | 820 | 689 | 1500 | 634 |
| 160 | 745.7 | 840 | 687.2 | 1520 | 632.5 |
| 180 | 744 | 860 | 685.5 | 1540 | 630.9 |
| 200 | 742.1 | 880 | 683.9 | 1560 | 629.4 |
| 220 | 740.2 | 900 | 682.4 | 1580 | 627.9 |
| 240 | 738.4 | 920 | 680.7 | 1600 | 626.4 |
| 260 | 736.8 | 940 | 679 | 1620 | 624.9 |
| 280 | 735 | 960 | 677.2 | 1640 | 623.3 |
| 300 | 733.4 | 980 | 675.6 | 1660 | 621.8 |
| 320 | 731.8 | 1000 | 674 | 1680 | 620.3 |
| 340 | 730 | 1020 | 672.4 | 1700 | 618.8 |
| 360 | 728.3 | 1040 | 670.8 | 1720 | 617.3 |
| 380 | 726.5 | 1060 | 669.1 | 1740 | 615.7 |
| 400 | 724.7 | 1080 | 667.5 | 1760 | 614.2 |
| 420 | 723 | 1100 | 666 | 1780 | 612.7 |
| 440 | 721.3 | 1120 | 664.3 | 1800 | 611.2 |
| 460 | 719.5 | 1140 | 662.6 | 1820 | 609.7 |
| 480 | 717.7 | 1160 | 661 | 1840 | 608.2 |
| 500 | 716 | 1180 | 659.3 | 1860 | 606.6 |
| 520 | 714.2 | 1200 | 657.9 | 1880 | 605.2 |
| 540 | 712.5 | 1220 | 656.4 | 1900 | 603.6 |
| 560 | 710.9 | 1240 | 654.8 | 1920 | 602.1 |
| 580 | 709.3 | 1260 | 653.2 | 1940 | 600.6 |
| 600 | 707.5 | 1280 | 651.6 | 1960 | 599 |
| 620 | 705.8 | 1300 | 650 | 1980 | 597.5 |
| 640 | 704.1 | 1320 | 648.3 | 2000 | 596 |
| 660 | 702.5 | 1340 | 646.7 |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |

