

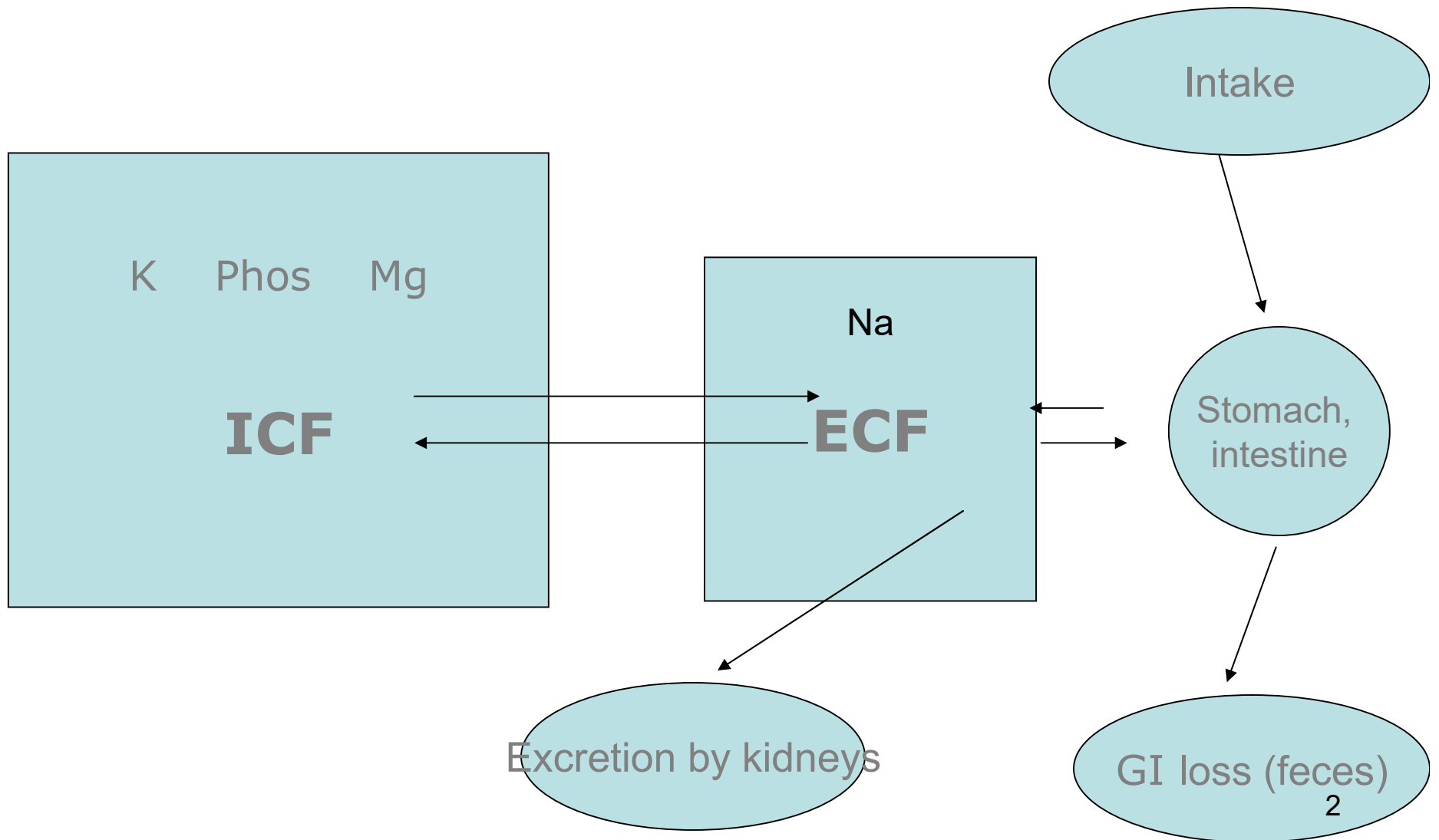
ELECTROLYTES

Extracellular fluid: Na, Cl ve HCO₃

Intracellular fluid: K, Mg, PO₄

Na,K pump: Maintains high levels of Na and K outside and inside the cell respectively.

Elimination of Intracellular ions



SODIUM:

Serum: 135-145 mmol/L

Excretion with urine: 30-280 mmol/day

Hyponatremia:

<135 mEq/L, (<120 clinical symptoms)

Neurological abnormalities, impaired nausea in mental responses, muscle cramps, dissociation, confusion, convulsions, coma.

Causes of hyponatremia:

It is the most common electrolyte

Increase of body water

Renal insufficiency

Heart failure

Cirrhosis

Decrease of body sodium

Diuretic therapy (thiazide)

Diabetic Ketonuria, Vomiting, diarrhea, sweating
burns, Uremia



Hyponatremia ("Water intoxication")

Water intoxication in the use of extacy (MDMA) Water intoxication in running athletes (serum Na <120 mmol / L)

($\frac{1}{4}$ teaspoon salt to 1 lt water).

Pseudo and Artifact hyponatremia

Increase of some substances in the serum results with changes in the water compartment- dilutional hyponatremia)

Hyperglycemia, (every 100 mg / dl glucose increase leads to 1.6 mmol / L decrease in serum Na)

Mannitol treatment (osmotic diuresis)

Hyperlipemia and hyperproteinemia can also result with hyponatremia.

Hypernatremia:

> 145 mEq / L

Edema, irritability, tremor, confusion, coma

Reasons for Hypernatremia: (loss of water or increase in relative Na)

Fever

Water loss (diuresis, excessive sweating and diarrhea)

Renal diseases,

Heart failure

Cirrhosis and ascit

Sodium (Na⁺)

- Normal S_{Na} : 135-145
- Major component of serum osmolality
 - $S_{osm} = (2 \times Na^+) + (BUN / 2.8) + (Glu / 18)$
 - Normal: 285-295
- Alterations in S_{Na} reflect an abnormal water regulation

$$\text{Serum Osmolality} = (2 \times (Na + K)) + (BUN / 2.8) + (glukose / 18) \\ (2 \times Na^+) + \dots\dots\dots$$

Case :

A patient with IDDM feels hypoglycemia symptoms in the morning and drinks 2 glasses of sugary drink. She has an appointment in the hospital and does not inject insulin with the scare of hypoglycemia shock.

Test results: Glucose 28 mmol/L (2.8-6.2 mmol/L)

Na 126 mmol/L (135-145 mmol/L)

K, Urea, etc are normal

Reasons for hyponatremia?

Hyponatremia results from dilution. Water shifts from intracellular to extracellular (intravascular) space to balance osmolality.

POTASSIUM: (neuromuscular and muscular irritability)

Plasma: 3.3-4.9 mmol/L,

Low or high K + causes muscle relaxation, irritability and paralysis, tachycardia, cardiac arrest

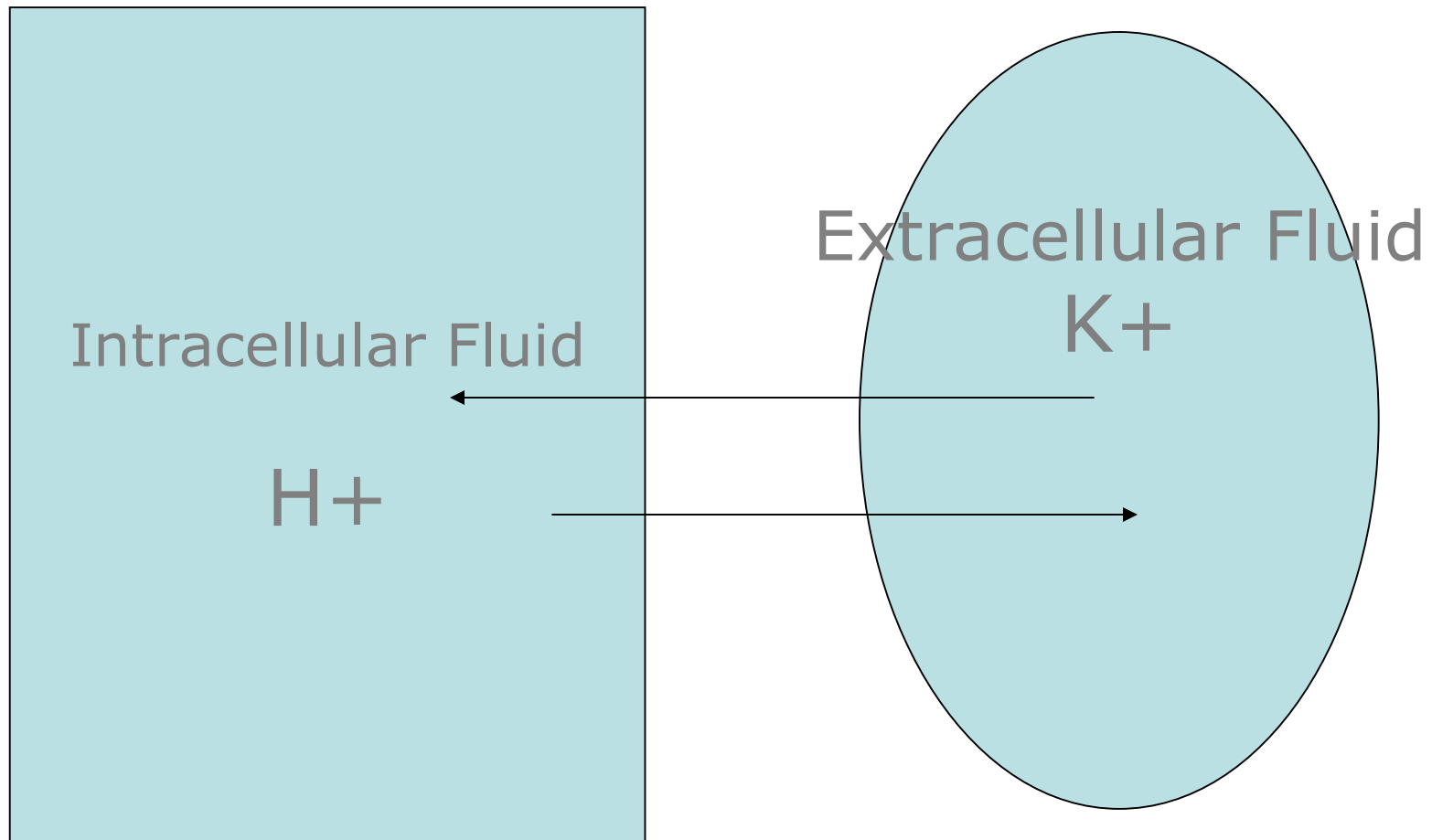
Hypokalemia:

Plasma potassium <3.0 mmol / l.

1. Reduction of dietary intake
2. Disruption of distribution in the body
 - insulin
 - Metabolic Alkalose
 - Dehydration

POTASSIUM

Metabolic Alkalosis and Hypokalemia



3. Urinary or GI losses

Kidney failure

Diarrhea, vomiting

Sweating

4. Drugs

a- Urinary loss: aminoglycosides, amphotericin B, corticosteroids, diuretics (thiazide and loop diuretics), levodopa, nifedipine, penicillin, rifampin

b- GI losses: laxatives

c- Redistribution in the body compartments:
Beta-2 agonists, lithium

Hyperkalemia:

Hemolysis

Metabolic acidosis

Dehydration

Insulin insufficiency

Kidney diseases

Sickle cell disease

rhabdomyolysis

burns

Blood transfusion

Drugs: diuretics lead to K⁺ retention

(Triamterene, Amiloride

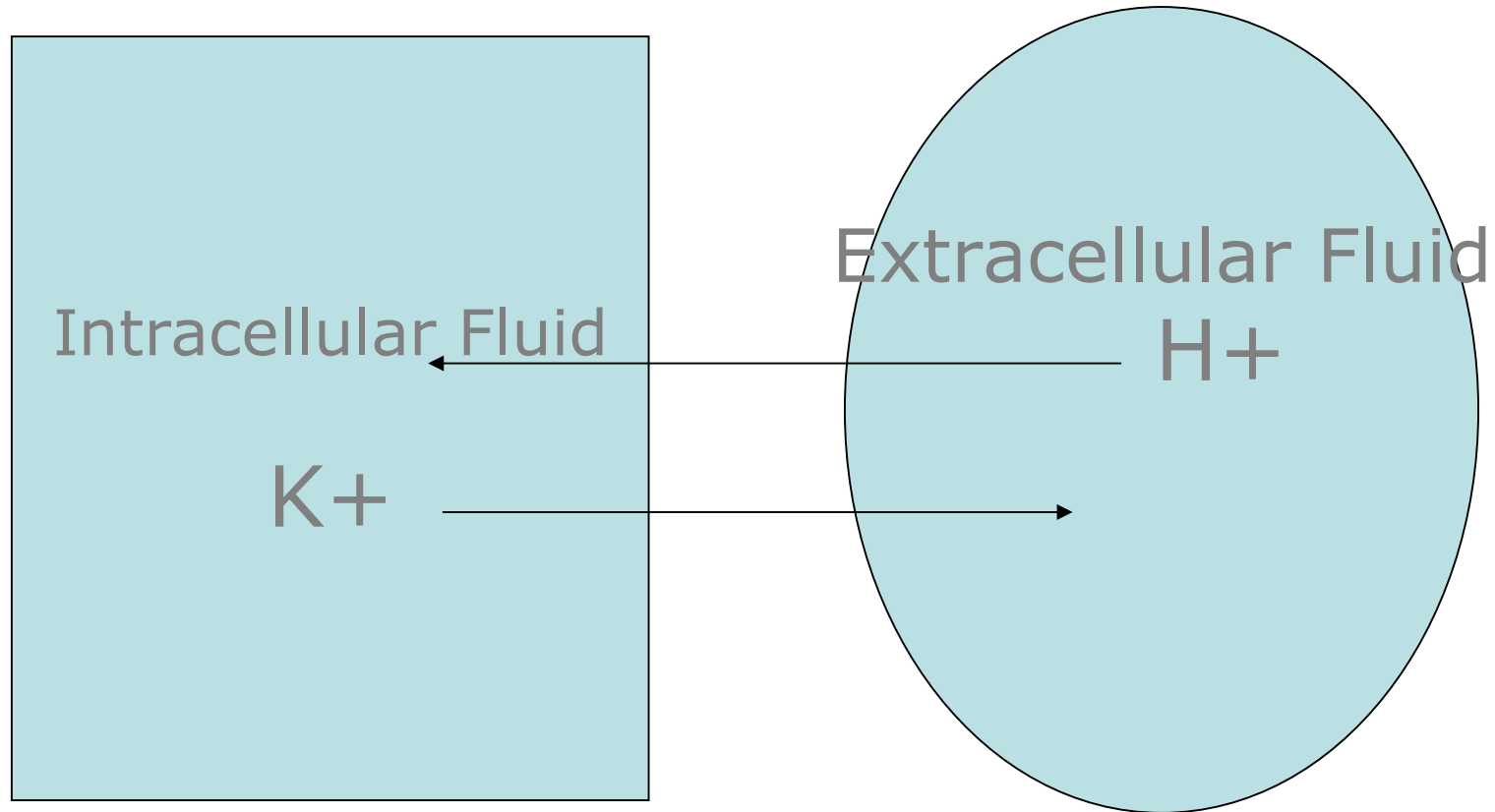
Spiro lactone)

Other medicines:

ACE inhibitors, Angiotensin receptor blockers, Digital toxicity, Beta blockers, High dose penicillin

POTASSIUM

Metabolic Acidosis and Hyperkalemia



Because acidosis causes increased H^+ in the cell, it directs K^+ to circulation. Each 0.1 unit decrease in pH 7.4 raises serum K^+ to 0.6 mEq / L.

Proteinuria:

Kidney disease

Effort proteinuria (Seen in adolescence)

DM

THANK YOU...