Stability Analyzes and Evaluations in Drug Formulations

**Prof. Dr. Asuman BOZKIR** 

#### **STABILITY**

The stability of a drug means that its production date characteristics during production, storage, transport, hospital and pharmacy storage and handling remain within predetermined limits.

# Shelf life

 It is the time period for which an active substance or pharmaceutical product is expected to conform to the specifications made by the stability studies performed on a certain number of batches when stored in the recommended package in the recommended condition.

## **Expiration date**

 The expiry date on an active substance or label of a pharmaceutical product is expected to remain within the specification limits of the substance or the product if stored under appropriate conditions.

- 1. Chemical stability
- 2. Physical stability
- 3. Microbiological stability
- 4. Therapeutic stability
- 5. Toxicological stability

A drug should also fulfill these 5 stability properties during its shelf life.

### Chemical stability and evaluations

- Oxidation reactions
- Hydrolysis reactions

### **Oxidation reactions**

- Examples are such as riboflavin, streptomycin, sulfadiazine, tetracycline, penicillins, novobiocin, norepinephrine, hydrocortisone, cyanocobalamin, chlorpromazine, ergometrine, A, B, C and E vitamins.
- $RH \rightarrow R. + H$  initial stage
- $R.+O_2 \rightarrow ROO.$
- $RH + RO_2$ .  $\rightarrow ROOH + R$ .
- R.+  $O_2 \rightarrow RO_2$  progression stage
- ROOH  $\rightarrow$  RO.+.OH

 $RO_2$ . +ROOH  $\longrightarrow$  Inactive products  $RO_2$ . +X \_\_\_\_Inactive products ROO + ROO --> Inactive products X:It is a free radical inhibitor. It comes from the auxiliary substances in the

environment.

End

- Self-initiated oxidation reactions without heat, light or heavy metal ions are called "auto-oxidation" reactions.
- Oxidation and reduction are reactions that move together.

Reduced state  $\leftarrow \rightarrow$  oxidized state+n  $e^{-}$ 

It is called "redox" or e transfer reaction.

Functional groups exposed to autoxidation;

Phenols, catechols, ethers, thiols, thioethers, carboxylic acids, nitrites, aldehydes, amines.