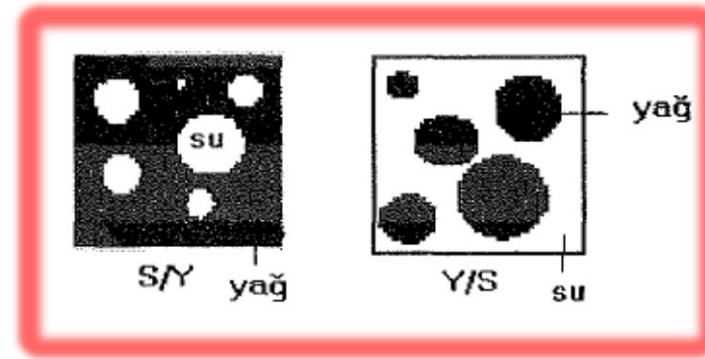


# Emulsions

- ➔ Emulsions are homogeneous heterogeneous dispersant systems formed by dispersing one or more of the at least two unmixed liquids in droplets with the help of an emulsifier.



## How should a stable emulsion be?

- During the shelf life, there should be no separation in the dispersed phase
- Redisperse should be able to return to its original state with a little rinsing and be easily homogenous
- The flow should be comfortable.

- Coalescence should not occur in internal phase in a stable pharmaceutical emulsion
- Cream should not form
- It should be able to maintain its fragrance, color, good looks and other physical properties.

**As a problem with instability in a pharmaceutical emulsion;**

**Flocculation and creaming,**

**Coalescence and phase separation (breaking)**

**Phase transformation**

**Various physical and chemical changes**

## **Three major factors influence flocculation:**

**Shaking intensity**

**Pushing force between droplets (Electrical loads  
between droplets pushing each other)**

**Van der Waals tug forces**

## **PHASE SEPERATION**

**Creaming should be considered different from phase separation;**

**Creation recycled**

**Phase decomposition is irreversible.**

- **Globules can not be dispersed by mixing when the film surrounding the droplets is broken.**

**Phase decomposition;**

**Droplet size**

**Viscosity of the dispersion medium**

**It depends on the phase volume ratio.**

**The most stable emulsion comes into play when the phase to volume ratio is 50/50.**

**It shows a more semi-solid state. In this respect, the dispersed phase means that the stability of the emulsion is increased as soon as it is close to this.**