




3. WEEK


PARTITION

COEFFICIENT

WHAT IS PARTITION COEFFICIENT?

- In its simplest form, "Partition Coefficient" is the expression of how the active substance diffuses between the solvent and water.


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- ❖ Passive absorbed active substances must pass biological membranes before entering the bloodstream.
 - ❖ Due to the lipoidal structure of the membranes, the rate of passage of the drug depends on the lipophilic nature of the active substance.

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- The lipophilic or hydrophilic character of an active substance can be determined by measuring the oil / water partition coefficient of the molecule.
 - In the measurement of the partition coefficient, "Octanol" or "Chloroform" is usually used as the oil phase.

- ❖ The most commonly used method for determining the partition coefficient is the "Shaking Method".
- ❖ At a certain temperature, a certain volume of water and octanol are added to a vessel to add an excess of active substance. It is determined how much active substance is present in the samples taken at certain time intervals.


$$\log K = \log [C_{\text{octanol}} / C_{\text{water}}]$$


- ❖ As the result of the experiment, mathematically mostly small numerical values are obtained, so the partition coefficient is logarithmically expressed.



❖ Non-ionic states of active substances have a higher partition coefficient, so they are absorptive in vivo even faster than membranes.


❖ There is also a linear relationship between the absorption rate and the degree of absorption of an active substance and the partition coefficient.

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- ❖ The main parameter required to estimate the absorption of an active substance in vivo at ambient conditions is the partition coefficient.
 - ❖ Another area where the partition coefficient has an effective role other than absorption is the solubility properties of the active substances.



❖ The role of the partition coefficient ($\log K$) values on absorption can be expressed numerically as follows.

- The $\log K$ value of the active substance is between 1 and 3, indicating that the active substance is absorptively elevated in the in vivo environment.
- If the active substance has a $\log K$ value greater than 6, the active substance has poor absorption.



❖ In this case the following interpretation can be made for the P value with the anti-logarithm being taken:

- Active ingredients below P value 1 are hydrophilic character,

- The active substances above the P value of 1 are Lipophilic characters.