

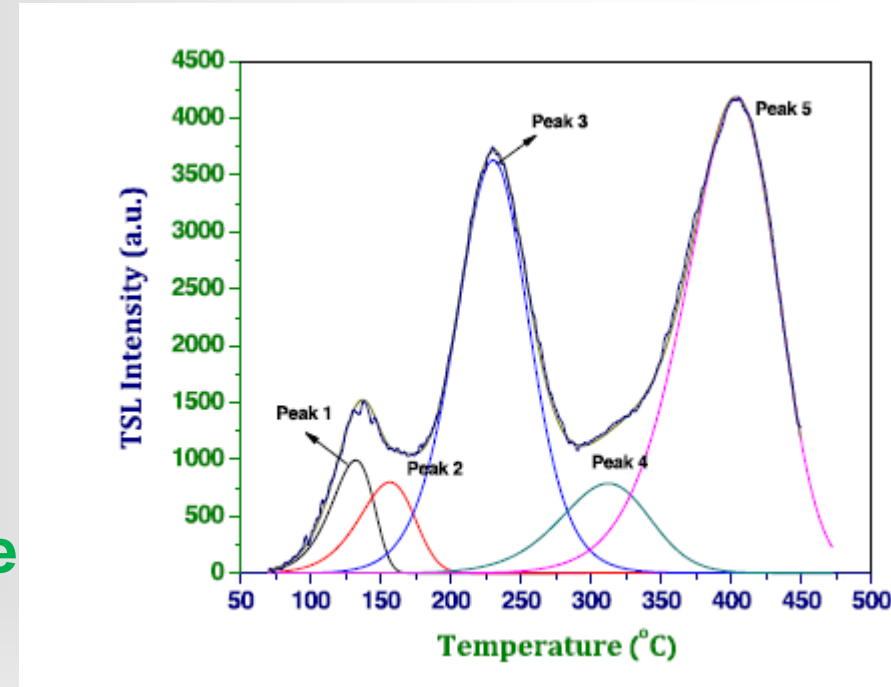
# Thermal Cleaning (peak separation)

## Overlapping peaks

→ sample is heated to appropriate incremental intermediate temperatures.

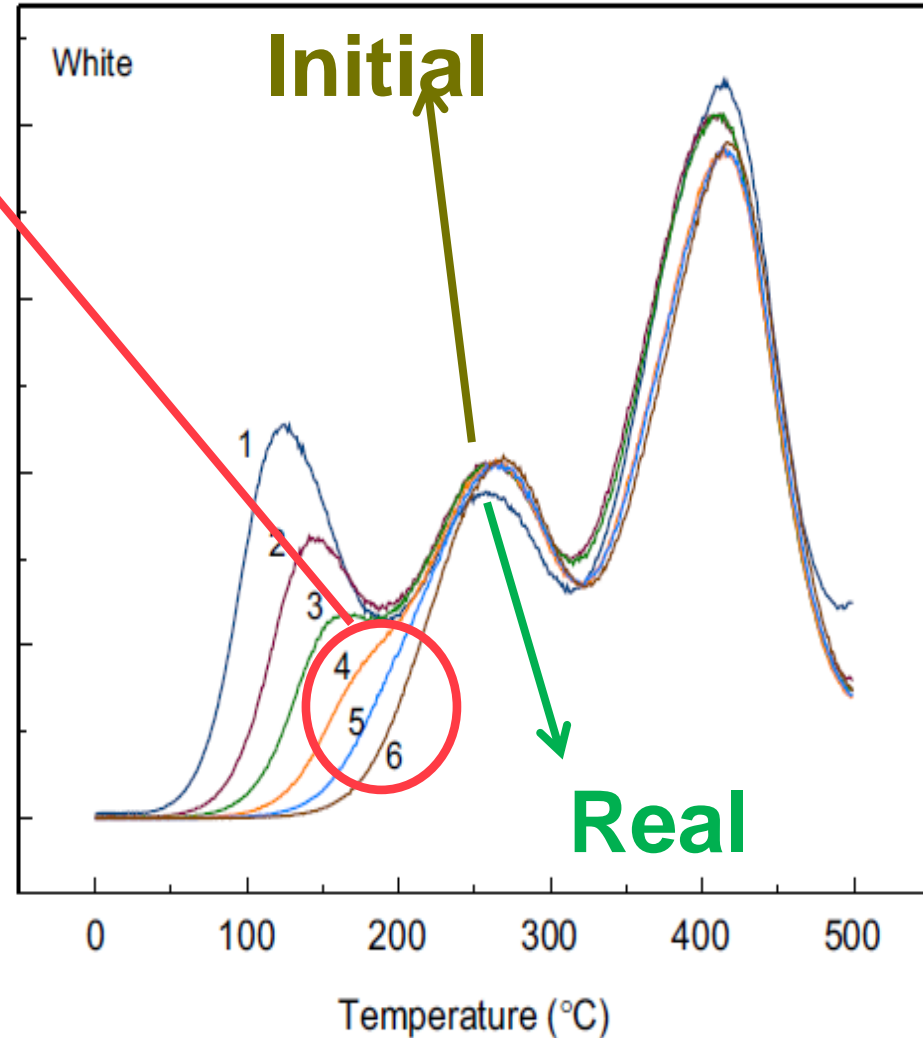
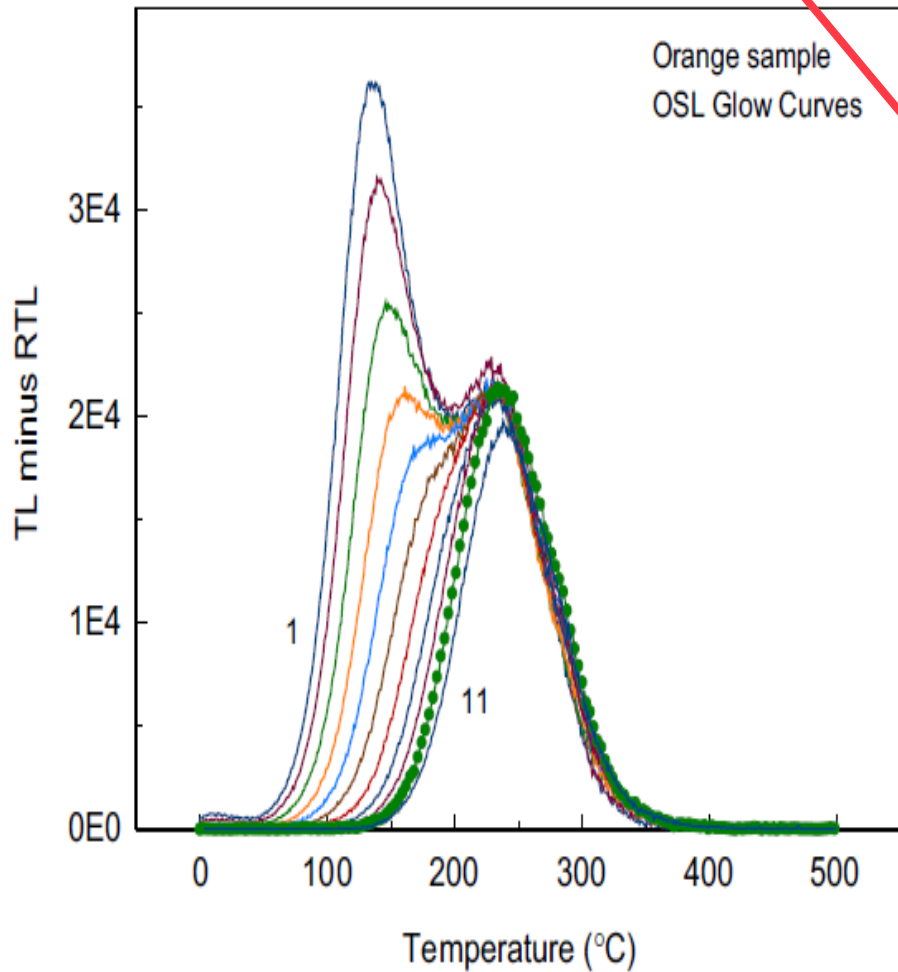
This technique is used when the energy activation  $E$  has to be determined by the initial rise method where a clean initial rise the peak is needed.

Thermal cleaning allows to determine the temperature of the maximum,  $T_{\max}$ , but not the intensity at the maximum because the cleaning thermal treatment could affect the number of the trapped charges.

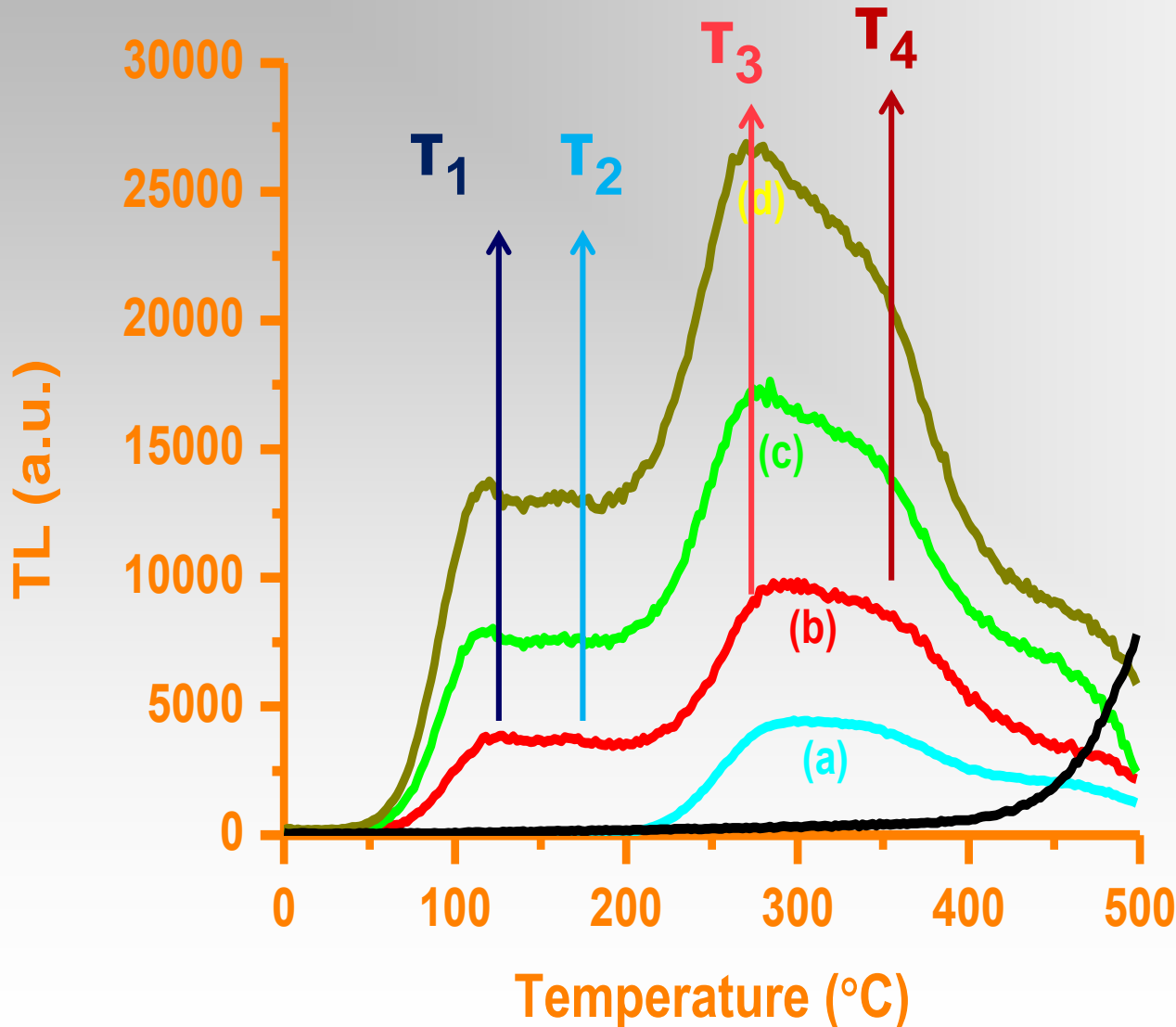


# Thermal Cleaning (peak separation)

Checking linearity of  $\ln(TL)$  vs  $1/kT$  plot



# Thermal fading



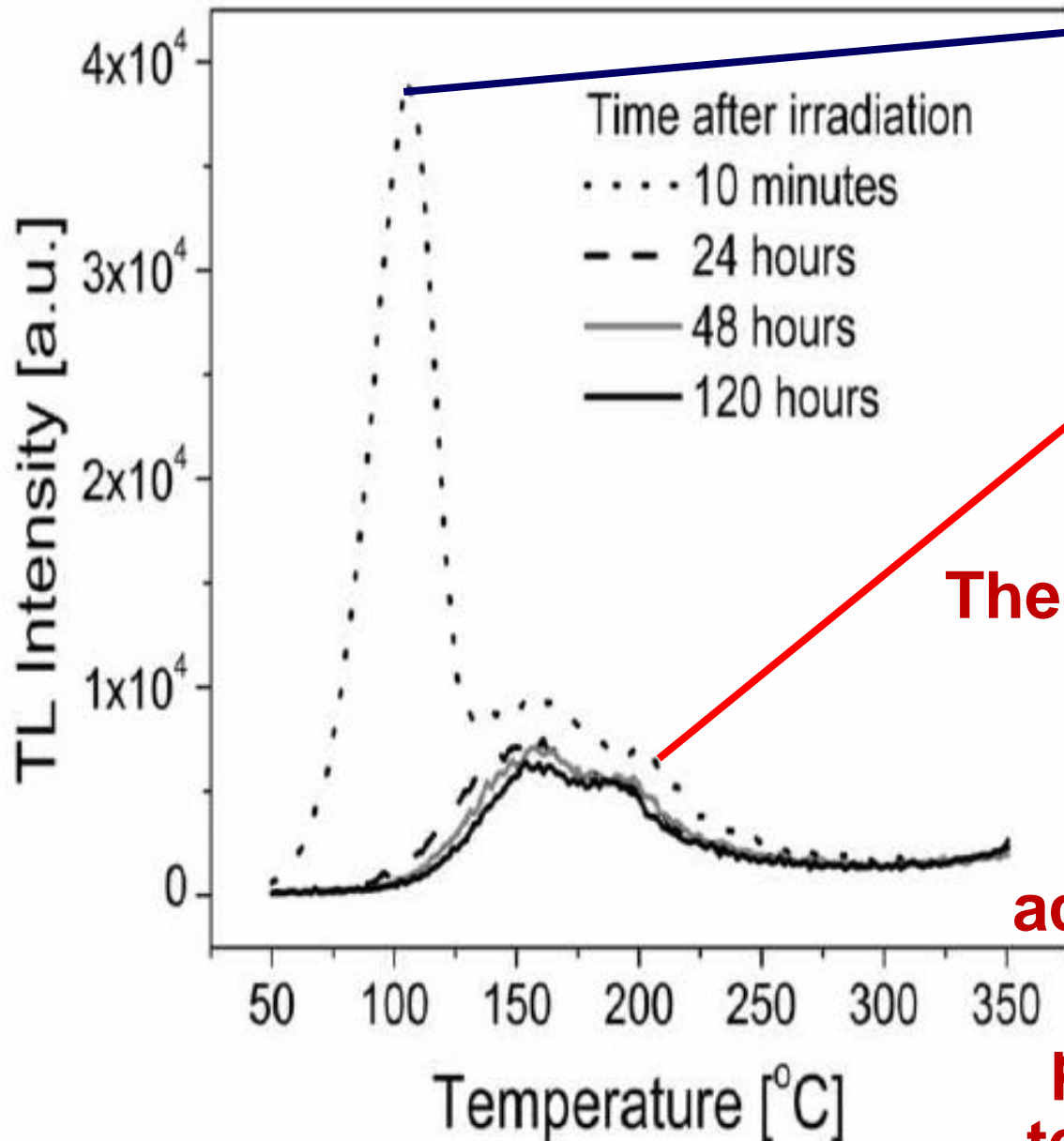
- $T_1 < T_2 < T_3 < T_4$
- The decrease of the TL peak's intensity according to the expected kinetic models and life times of traps

# Anomalous fading

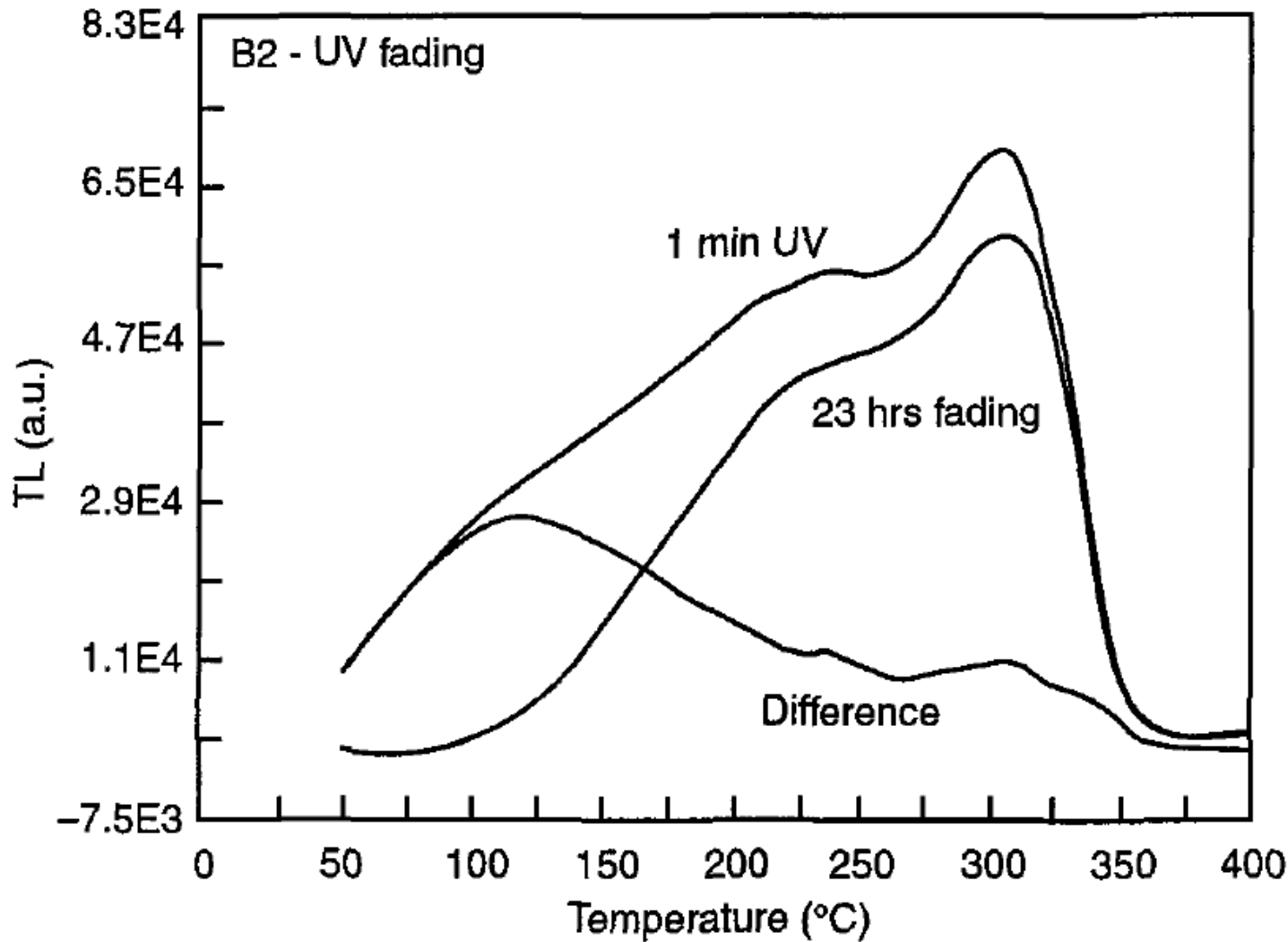
$\tau_1 \sim 1-2$   
hours

$\tau_2 \sim 5$  years

The charges are released by the trap at a rate which is much faster than those expected according to the lifetime of the trap while the phenomenon is weakly temperature dependent

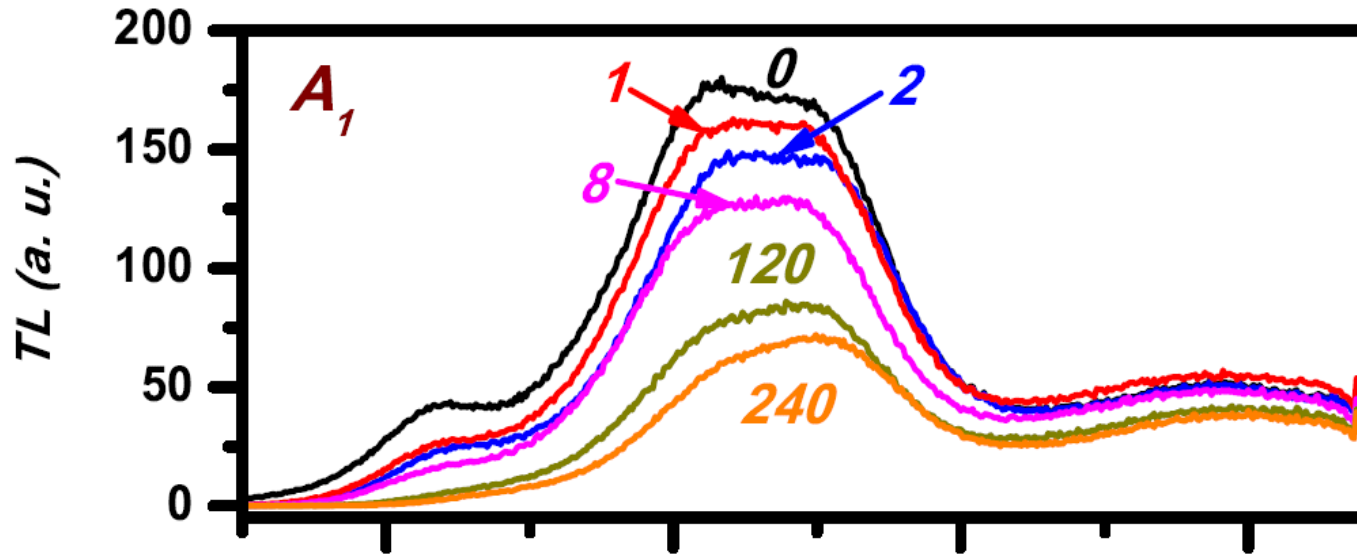


# Anomalous fading: examples

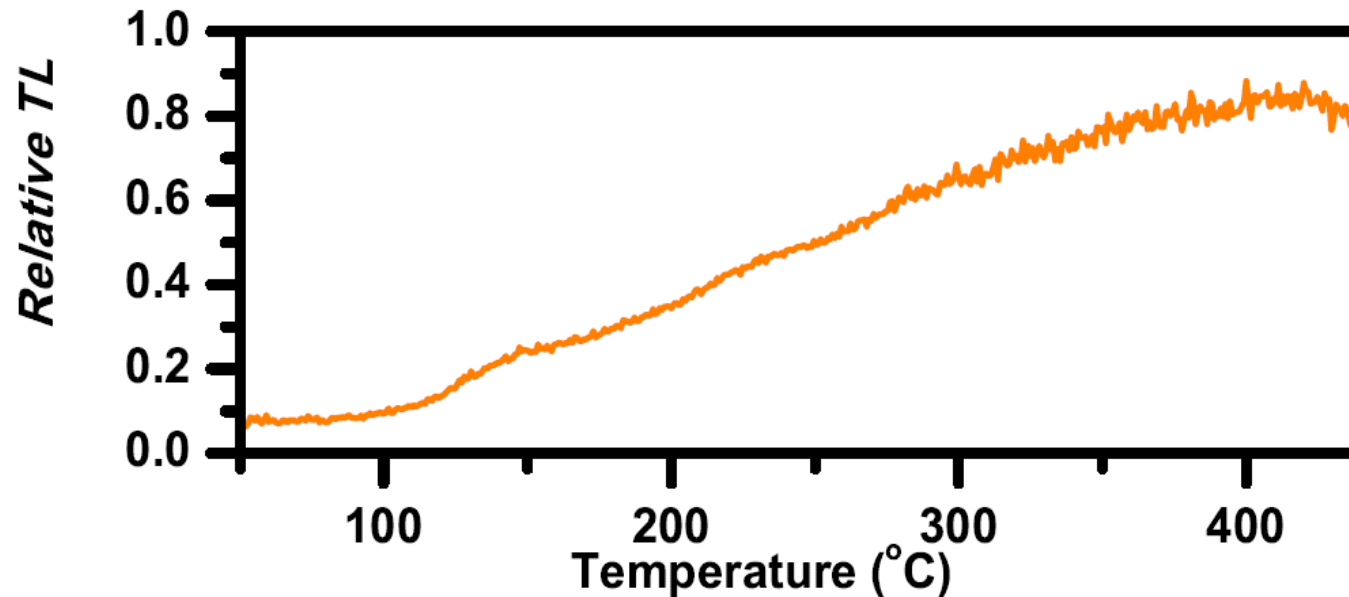


**CVD  
diamond**

# Anomalous fading: examples

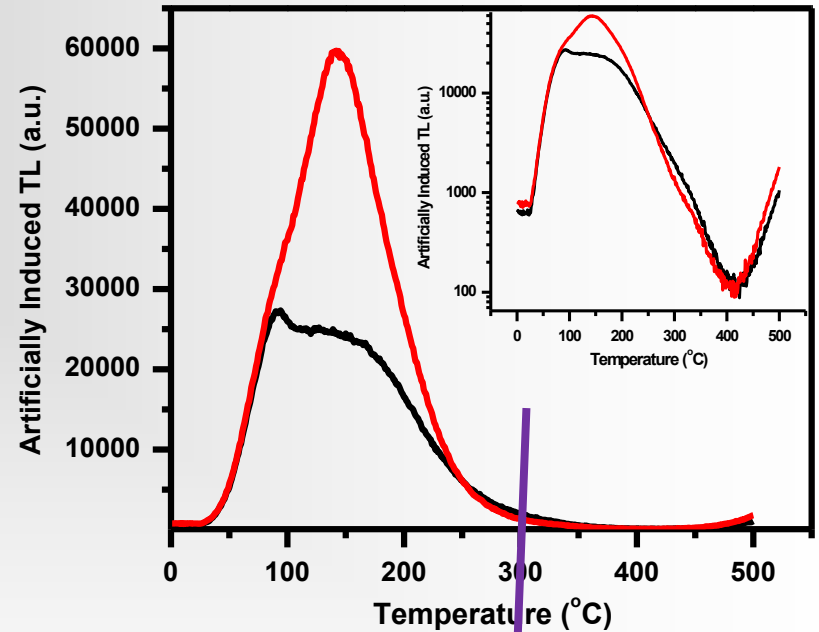
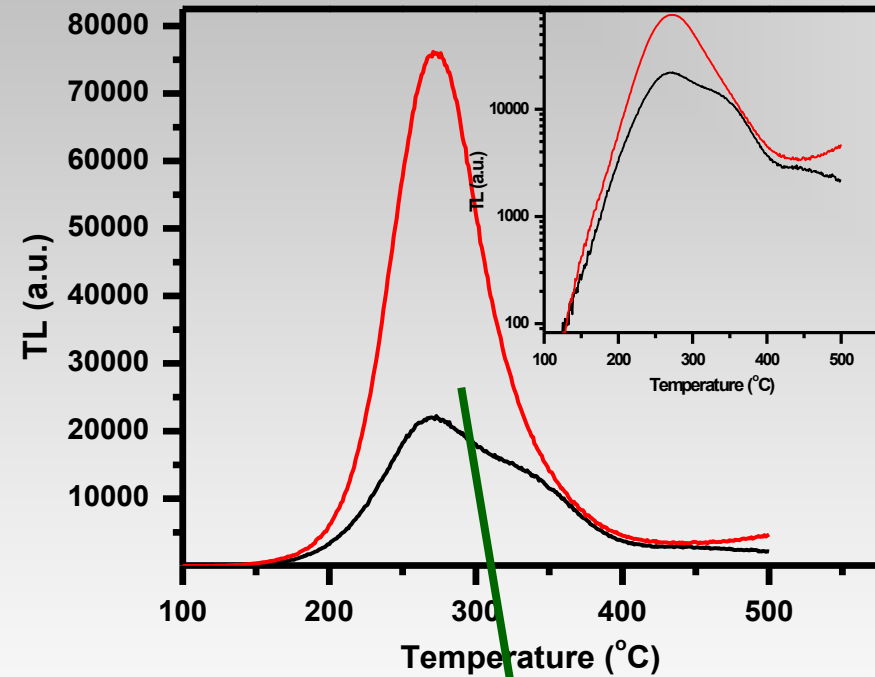


Fluor-apatite  
 $\text{Ca}_5(\text{PO}_4)_3\text{F}$



The anomalous fading is observed in natural minerals, as well as in TL materials as ZnS:Cu, ZnS:Co, CaF<sub>2</sub>:Mn, KCl:Tl, etc.

But mostly feldspars ( $x[\text{AlSi}_3\text{O}_8]$ ,  $x = \text{K, Na, Ca}$ )



## Tunnelling

An electron trapped in a level A of an atom (Fig.6) may recombine directly with a hole in a level B of another atom without involving the delocalized bands. Mikhailov gave in 1971 a model for tunnelling process [1-3].

The defects responsible for levels A and B must be closed to each other. This can occur when traps and recombination centers are in a very high concentration and when the two centers belong to the same defect site. This transition occurs through the potential barrier (tunnelling) which separates the electron in A from the hole in B. The recombination results in the emission of luminescence. The effect is athermal.

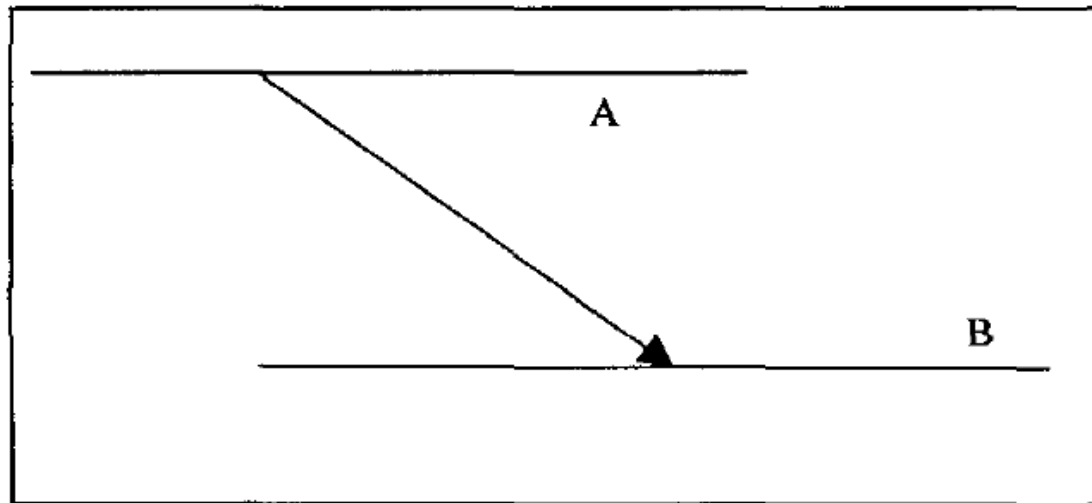
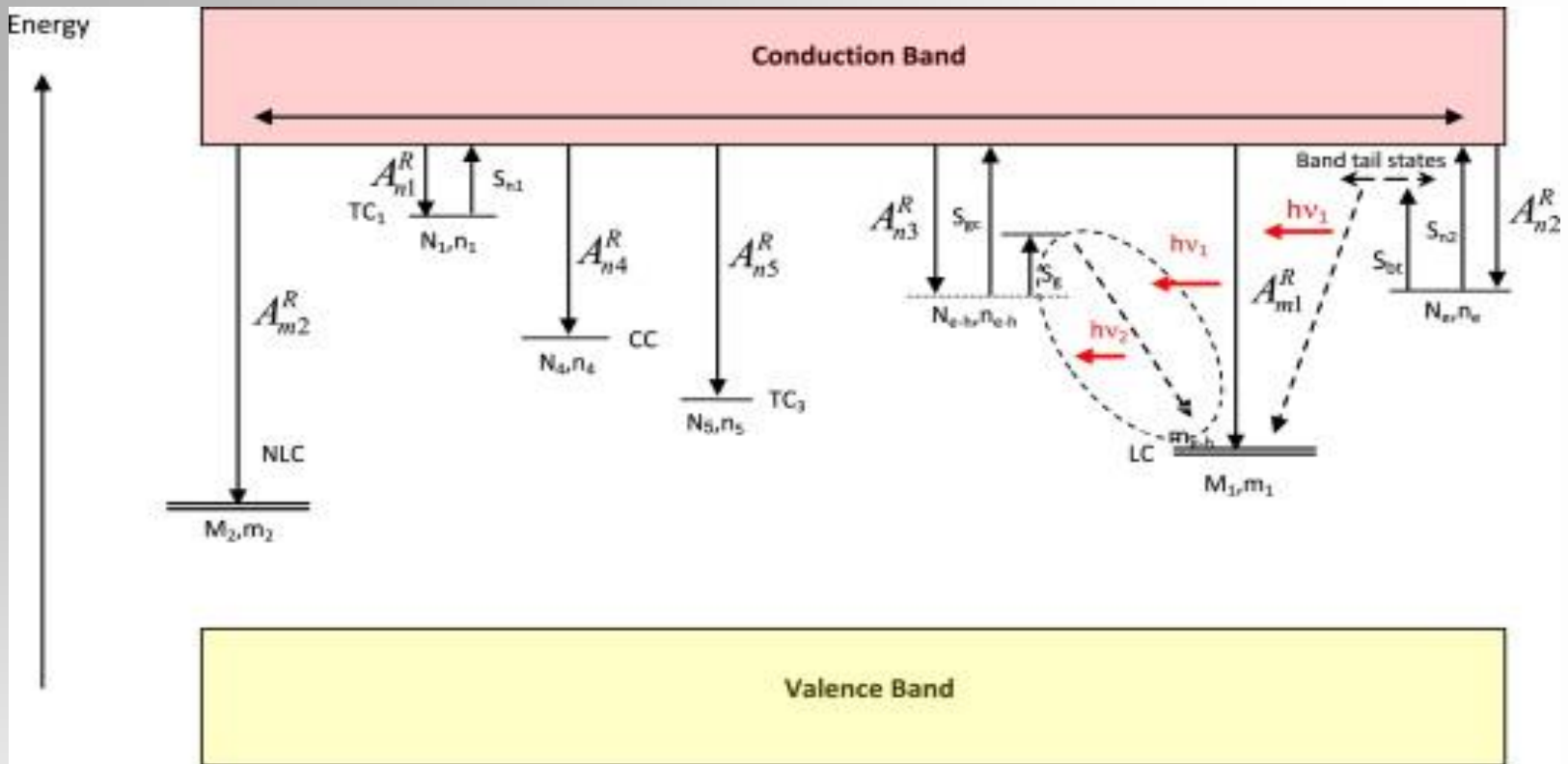


Fig.6. Tunnelling between an electron in A and a hole in B.





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**Nanodosimetric kinetic model incorporating localized and delocalized recombination: Application to the prediction of the electron dose response of the peak 5a/5 ratio in the glow curve of LiF:Mg,Ti (TLD-100)**

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■ <http://dx.doi.org/10.1016/j.radmeas.2014.01.015>